



EDEXCEL INTERNATIONAL GCSE (9–1)

# SCIENCE DOUBLE AWARD

Student Book

Brian Arnold, Phil Bradfield, Jim Clark, Penny Johnson, Steve Owen, Steve Potter,  
Steve Woolley, Rachel Yu

 **Pearson**



PEARSON EDEXCEL INTERNATIONAL  
GCSE (9–1)

# SCIENCE DOUBLE AWARD

Student Book

## BIOLOGY

Phil Bradfield  
Steve Potter

## CHEMISTRY

Jim Clark  
Steve Owen  
Rachel Yu

## PHYSICS

Brian Arnold  
Penny Johnson  
Steve Woolley

Published by Pearson Education Limited, 80 Strand, London, WC2R 0RL.

www.pearsonglobalschools.com

Copies of official specifications for all Edexcel qualifications may be found on the website: <https://qualifications.pearson.com>

Text © Pearson Education Limited 2017

Edited by Stephen Cunningham

Designed by Cobalt id

Typeset by Tech-Set Ltd, Gateshead, UK

Original illustrations © Pearson Education Limited 2017

Illustrated by Tech-Set Ltd, Gateshead, UK

Cover design by Pearson Education Limited

Cover photo/illustration © Getty Images: Jeffrey Rotman

Picture research by: Frances Topp, Andreas Schindler and Penny Bowden.

The rights of Phil Bradfield, Steve Potter, Jim Clark, Steve Owen, Rachel Yu, Brian Arnold, Penny Johnson and Steve Woolley to be identified as authors of this work have been asserted by them in accordance with the Copyright, Designs and Patents Act 1988.

First published 2017

20 19 18 17

10 9 8 7 6 5 4 3

British Library Cataloguing in Publication Data

A catalogue record for this book is available from the British Library

ISBN 9780435185282

### Copyright notice

All rights reserved. No part of this publication may be reproduced in any form or by any means (including photocopying or storing it in any medium by electronic means and whether or not transiently or incidentally to some other use of this publication) without the written permission of the copyright owner, except in accordance with the provisions of the Copyright, Designs and Patents Act 1988 or under the terms of a licence issued by the Copyright Licensing Agency, Barnard's Inn, 86 Fetter Lane, London, EC4A 1EN ([www.cla.co.uk](http://www.cla.co.uk)). Applications for the copyright owner's written permission should be addressed to the publisher.

Printed in Slovakia by Neografia

### Endorsement Statement

In order to ensure that this resource offers high-quality support for the associated Pearson qualification, it has been through a review process by the awarding body. This process confirms that this resource fully covers the teaching and learning content of the specification or part of a specification at which it is aimed. It also confirms that it demonstrates an appropriate balance between the development of subject skills, knowledge and understanding, in addition to preparation for assessment.

Endorsement does not cover any guidance on assessment activities or processes (e.g. practice questions or advice on how to answer assessment questions), included in the resource nor does it prescribe any particular approach to the teaching or delivery of a related course.

While the publishers have made every attempt to ensure that advice on the qualification and its assessment is accurate, the official specification and associated assessment guidance materials are the only authoritative source of information and should always be referred to for definitive guidance.

Pearson examiners have not contributed to any sections in this resource relevant to examination papers for which they have responsibility.

Examiners will not use endorsed resources as a source of material for any assessment set by Pearson. Endorsement of a resource does not mean that the resource is required to achieve this Pearson qualification, nor does it mean that it is the only suitable material available to support the qualification, and any resource lists produced by the awarding body shall include this and other appropriate resources.

### Picture Credits

The authors and publisher would like to thank the following individuals and organisations for permission to reproduce photographs:

(Key: b-bottom; c-centre; l-left; r-right; t-top)

**123RF.com:** 163bl, 378t, 381r, 326, choneschones 533, cristimatei 698tl, Feliks Gurevich 550bl, homestudio 543, Eric Isselee 230bl, Jozsef Szasz-Fabian 495tr, leungchopan 52b, Melinda Nagy 504tc, Scandal 46b; **Alamy Stock Photo:** age fotostock 514r, 666, Ageev Rostislav 600, Alison Eckett 650, Nathan Allred 22b, Ange 679r, Archive Pics 669l, Art Directors & TRIP 695, Aurora 456, Authentic Creations 582tl, Blend-Memento 196, blickwinkel 3cl, 3cr, 3bl, 3br, 25l, 230c, Bon Appetit 476, Nigel Cattlin 27l, 229, Charles Stirling 504tl, Chris Cooper-Smith 509tl, Chris Rose 572tr, Trevor Chriss 322, 529, David Colbran 218, Ashley Cooper 481r, Cultura Creative 381l, Cultura Creative (RF) 652, Design Pics Inc 621r, Digital Image Library 613c, Dorling Kindersley ltd 239tl, Emmanuel Lacoste 624l, Hayley Evans 22tr, FineArt 223, Flake 530, frans lemmens 602tr, GL Archive 669r, Granger Historical Picture Archive 520, Henry Westheim Photography 591tl, Horizon International Images Limited 601, 697, Rachel Husband 22tl, IanDagnall

Computing 23tc, Image Source 545, ImagineThat 573, INTERFOTO 625, Jeff Rotman 621c, JLMimages 24tl, John Joannides 597, Juniors Bildarchiv GmbH 230tr, LGPL / Ian Cartwright 378b, Linda Richards 508tr, Andrew Michael 559t, Michele Burgess 489tr, NASA 701, NASA / S.Dupuis 412, NASA Archive 523t, NASA Photo 488, National Geographic Creative 71, The Natural History Museum 629, Naturfoto-Online 225, Nick Greening 489tl, Sergey Nivens 236, paul ridsdale pictures 639, Nikos Pavlakis 366r, philipus 552, Pictorial Press Ltd 211, 608, 630t, 656, pixel shepherd 533, PjrStudio 572bl, Pulsar Images 684l, Radoslav Radev 582bl, Richard Wainscoat 698b, Rodney\_X 145cr, samart boonyang 646, sciencephotos 549c, 550c, 565, Scott Ramsey 504tr, Sean Pavone 591, Trevor Smith 95, Studioshots 533b, David Taylor 401l, Trevor Chriss 322, 529, Genevieve Vallee 145cl, Jan Wlodarczyk 163br, World History Archive 282, Zoonar GmbH 162, 482b, Zoonar GmbH 162, 482b, Владимир Галкин 495tc; **Fotolia.com:** annavaczi 602tl, artush 518, bit24 48, Grinchh 269, izzy71 602bl, Jess8 230br, Dr\_Kateryna 51, Kateryna\_Kon 198, ktsdesign 568, Arpad Nagy-Bagoly 478c, nengredeye 131, schankz 624r, troniphoto 509bl, Pavla ZAKOVA 230tl; **Getty Images:** Aleksandrs Podskocijs / EyeEm 598, Thomas Barwick 120, BlackJack3D 34, Carlos Herrera / Icon Sportswire / Corbis 591c, Graiki 522, In Pictures / Corbis / Gideon Mendel 425, Muditha Madushan / EyeEm 2, Miguel Malo 366l, Maxim Grigoryev 344, mevans 507bl, NASA / National Geographic 258, PaulFleet 414, Peter Turnley 514tl, qbanczyk 93r, ROBERT SULLIVAN / AFP 572c, Science & Society Picture Library 613b, Scott Eells / Bloomberg 575b, ViewStock 507tl, YinYang 508tl; **Maritime & Coastguard Agency:** 571r; **Nature Picture Library:** Nature Production 436t; **Pearson Education Ltd:** Studio 8 541, Gareth Boden 549tl, Coleman Yuen. Pearson Education Asia Ltd 530bl, Oxford Designers & Illustrators Ltd 481l, Jules Selmes 538, 602br, Jules Selmes 538, 602br, Tsz-zhan Kwok 531; **Science Photo Library Ltd:** 73, 123, 352b, 393t, 401c, 401r, 436b, AJ PHOTO 42, Andrew Lambert Photography 152, 272, 306, 309, 352t, 354r, 362t, 363, 403t, 403b, 498, 511, 630b, Animated Healthcare Ltd 104l, STEPHEN AUSMUS / US DEPARTMENT OF AGRICULTURE 23tl, Biomedical Imaging Unit, Southampton General Hospital 19, BIOPHOTO ASSOCIATES 43, 52, 53, 72, 124, Robert Brook 188, Carol and Mike Werner 446, Martyn F Chillmaid 54b, 165, 296, 345, 346, 348, 362b, 376, 384, 388, 390, 393l, 400, 401b, 401br, 402, 413, 420, 438b, 475, 478l, 565r, CNRI 226, 677, Crown Copyright / Health and Safety Laboratory 398, David Parker 179, 611b, Phil Degginger 415, Alan L. Detrick 180, Dorling Kindersley / UIG 24bl, 599, Dr Brad Mogan / Visuals Unlimited 250, DR GOPAL MURTI 6t, 25r, John Durham 6b, ESA - A. Le Floc'h 523c, MAURO FERMARIELLO 50b, Fundamental Photos 370, Gastrolab 60, GIPhotoStock 54t, 382, 383, 395, 401cr, 428cl, Dr Gopal Murti 6t, 25r, Steve Gschmeissner 139cr, 142, 203, Gusto Images 262, Adam Hart-Davis 146, RALPH HUTCHINGS, VISUALS UNLIMITED 36, Mikkel Juul Jensen 316, Keith R. Porter 5, Leonard Lessin 78t, 78b, DR P. MARAZZI 678, Microscope 111, CORDELIA MOLLOY 506tl, 540tl, 540tc, National Library of Medicine 222, NATURAL HISTORY MUSEUM, LONDON 224t, 224b, NIBSC 37, Susumu Nishinaga 139cl, Omikron 104r, David Parker 179, 611b, Alfred Pasiaka 455, Lea Paterson 447l, PATRICK LANDMANN 684r, POWER AND SYRED 23b, R. MEGNA / FUNDAMENTAL PHOTOS 537, JC Revy, ISM 139tl, Saturn Stills 94, Science Pictures Ltd 204, Martin Shields 145b, Lee D. Simon 27c, Sinclair Stammers 25c, Volker Steger 478r, Steve Gschmeissner 139cr, 142, 203, David Taylor 401cl, TONY & DAPHNE HALLAS 611t, Trevor Clifford Photography 136, UIG / Dorking Kindersley 433, Charles D Winters 330, 353, 354l, 482t; **Shutterstock.com:** 279, photo Studio 178, 805084 590, 463c, 573r, 463c, 573r, Africa Studio 50t, Aivolie 298, Anastasios71 358, Vartanov Anatoly 572tl, Galyna Andrushko 386, arka38 239br, H E Benson 259l, best images. 462b, bikeriderlondon 571l, Bitt24 473, Sinisa Botas 462t, Brian Kinney 613t, Sherri R. Camp 428tl, Cardens Design 694, Norman Chan 542, Chechubis 438t, Marcel Clemens 288b, Dashu 459, Demarcomedia 287, Dencg 266l, devy 495tl, Digieva 239tr, Dr. OGA 447c, EpicStockMedia 558, Peter Hermes Furian 323, Mauricio Graiki 93l, Jiri Hera 335bl, IANG HONGYAN 251, imagedb.com 532, Joanne Harris and Daniel Bubnich 583, Anan Kaewkhammul 460, karrapavan 638, KDEdesign 335tl, Jatuporn Khuansuwan 308, Kateryna Kon 26t, Ktsdesign 26b, 290, Robyn Mackenzie 121, Ilya Malov 428cr, MidoSemsem 582c, MIGUEL GARCIA SAAVEDRA 530br, Minerva Studio 627b, morchella 584, Resul Muslu 239bl, Mylisa 266r, Romanova Natali 276r, Oleg1969 276l, Olga Selyutina 559b, Palmaria 483, Pavel L Photo and Video 575t, Werayuth Priyapornprapa 463t, Przemyslaw Skibinski 577, ra3rn 644, Vadim Ratnikov 463b, Remedios55 335br, Sander van Sinttruye 702l, Aygul Sarvarova 481, Scanrail1 336, Jeff Schultes 428tr, sciencepics 79, SFC 564, Smileus 562, Sozajiten 23tr, Sam Strickler 447r, STUDIOMAX 603c, Mary Terribery 447, Tiago Ladeira 605, Reinhard Tiburzy 241, topseller 621l, Dave Turner 347, Valeriy Lebedev 702bl, VanHart. 259r, Ventin 288t, Volodymyr Goinyk 620, Voyagerix 506bl, Wolfgang Kloehr 702r, XXLPhoto 464t, YanLev 491, Ron Zmiri 377.

**Inside front cover: Shutterstock.com:** Dmitry Lobanov

All other images © Pearson Education

**Disclaimer: neither Pearson, Edexcel nor the authors take responsibility for the safety of any activity.** Before doing any practical activity you are legally required to carry out your own risk assessment. In particular, any local rules issued by your employer must be obeyed, regardless of what is recommended in this resource. Where students are required to write their own risk assessments they must always be checked by the teacher and revised, as necessary, to cover any issues the students may have overlooked. The teacher should always have the final control as to how the practical is conducted.

<b>COURSE STRUCTURE</b>	<b>iv</b>	<b>UNIT 3</b>	<b>558</b>
<b>ABOUT THIS BOOK</b>	<b>vi</b>	<b>UNIT 4</b>	<b>590</b>
<b>ASSESSMENT OVERVIEW</b>	<b>viii</b>	<b>UNIT 5</b>	<b>620</b>
<b>BIOLOGY</b>		<b>UNIT 6</b>	<b>638</b>
<b>UNIT 1</b>	<b>2</b>	<b>UNIT 7</b>	<b>656</b>
<b>UNIT 2</b>	<b>34</b>	<b>UNIT 8</b>	<b>694</b>
<b>UNIT 3</b>	<b>120</b>	<b>APPENDICES</b>	
<b>UNIT 4</b>	<b>162</b>	<b>APPENDIX A: PERIODIC TABLE</b>	<b>708</b>
<b>UNIT 5</b>	<b>196</b>	<b>APPENDIX B: COMMAND WORDS</b>	<b>709</b>
<b>UNIT 6</b>	<b>236</b>	<b>INDEX</b>	<b>710</b>
<b>CHEMISTRY</b>		<b>ON THE EBOOK</b>	
<b>UNIT 1</b>	<b>258</b>	<b>APPENDIX C: PHYSICAL FORMULAE</b>	<b>720</b>
<b>UNIT 2</b>	<b>344</b>	<b>APPENDIX D: PHYSICAL QUANTITIES</b>	<b>721</b>
<b>UNIT 3</b>	<b>412</b>	<b>APPENDIX E: EXPERIMENTAL AND INVESTIGATIVE SKILLS</b>	<b>722</b>
<b>UNIT 4</b>	<b>446</b>	<b>BIOLOGY GLOSSARY</b>	<b>731</b>
<b>PHYSICS</b>		<b>CHEMISTRY GLOSSARY</b>	<b>741</b>
<b>UNIT 1</b>	<b>488</b>	<b>PHYSICS GLOSSARY</b>	<b>747</b>
<b>UNIT 2</b>	<b>528</b>		

# BIOLOGY

## UNIT 1: ORGANISMS AND LIFE PROCESSES

- |   |                                 |    |
|---|---------------------------------|----|
| 1 | LIFE PROCESSES                  | 3  |
| 2 | THE VARIETY OF LIVING ORGANISMS | 22 |

## UNIT 2: ANIMAL PHYSIOLOGY

- |   |                            |     |
|---|----------------------------|-----|
| 3 | BREATHING AND GAS EXCHANGE | 35  |
| 4 | FOOD AND DIGESTION         | 48  |
| 5 | BLOOD AND CIRCULATION      | 64  |
| 6 | COORDINATION               | 77  |
| 7 | CHEMICAL COORDINATION      | 91  |
| 8 | HOMEOSTASIS AND EXCRETION  | 97  |
| 9 | REPRODUCTION IN HUMANS     | 104 |

## UNIT 3: PLANT PHYSIOLOGY

- |    |                                 |     |
|----|---------------------------------|-----|
| 10 | PLANTS AND FOOD                 | 121 |
| 11 | TRANSPORT IN PLANTS             | 136 |
| 12 | CHEMICAL COORDINATION IN PLANTS | 145 |
| 13 | REPRODUCTION IN PLANTS          | 151 |

## UNIT 4: ECOLOGY AND THE ENVIRONMENT

- |    |                                     |     |
|----|-------------------------------------|-----|
| 14 | ECOSYSTEMS                          | 163 |
| 15 | HUMAN INFLUENCES ON THE ENVIRONMENT | 176 |

## UNIT 5: VARIATION AND SELECTION

- |    |   |     |
|----|---|-----|
| 16 | CHROMOSOMES, GENES AND DNA                          | 197 |
| 17 | CELL DIVISION                                       | 202 |
| 18 | GENES AND INHERITANCE                               | 211 |
| 19 | NATURAL SELECTION, EVOLUTION AND SELECTIVE BREEDING | 222 |

## UNIT 6: MICROORGANISMS AND GENETIC MODIFICATION

- |    |                      |     |
|----|----------------------|-----|
| 20 | USING MICROORGANISMS | 237 |
| 21 | GENETIC MODIFICATION | 245 |

# CHEMISTRY

## UNIT 1: PRINCIPLES OF CHEMISTRY

- |   |   |     |
|---|---|-----|
| 1 | STATES OF MATTER                              | 259 |
| 2 | ELEMENTS, COMPOUNDS AND MIXTURES              | 266 |
| 3 | ATOMIC STRUCTURE                              | 276 |
| 4 | THE PERIODIC TABLE                            | 282 |
| 5 | CHEMICAL FORMULAE, EQUATIONS AND CALCULATIONS | 290 |
| 6 | IONIC BONDING                                 | 316 |
| 7 | COVALENT BONDING                              | 326 |

## UNIT 2: INORGANIC CHEMISTRY

- |    |                                    |     |
|----|------------------------------------|-----|
| 8  | THE ALKALI METALS                  | 345 |
| 9  | THE HALOGENS                       | 352 |
| 10 | GASES IN THE ATMOSPHERE            | 358 |
| 11 | REACTIVITY SERIES                  | 366 |
| 12 | ACIDS AND ALKALIS                  | 381 |
| 13 | ACIDS, BASES AND SALT PREPARATIONS | 386 |
| 14 | CHEMICAL TESTS                     | 398 |

## UNIT 3: PHYSICAL CHEMISTRY

- |    |  |     |
|----|--|-----|
| 15 | ENERGETICS                                 | 413 |
| 16 | RATES OF REACTION AND REVERSIBLE REACTIONS | 428 |

## UNIT 4: ORGANIC CHEMISTRY

- |    |                                   |     |
|----|-----------------------------------|-----|
| 17 | INTRODUCTION TO ORGANIC CHEMISTRY | 447 |
| 18 | CRUDE OIL                         | 459 |
| 19 | ALKANES                           | 468 |
| 20 | ALKENES                           | 473 |
| 21 | SYNTHETIC POLYMERS                | 478 |

# PHYSICS

## UNIT 1: FORCES AND MOTION

1	MOVEMENT AND POSITION	489
2	FORCES AND SHAPE	504
3	FORCES AND MOVEMENT	514

## UNIT 2: ELECTRICITY

4	MAINS ELECTRICITY	529
5	CURRENT AND VOLTAGE IN CIRCUITS	537
6	ELECTRICAL RESISTANCE	545

## UNIT 3: WAVES

7	PROPERTIES OF WAVES	559
8	THE ELECTROMAGNETIC SPECTRUM	568
9	LIGHT AND SOUND WAVES	575

## UNIT 4: ENERGY RESOURCES AND ENERGY TRANSFER

10	ENERGY TRANSFERS	591
11	THERMAL ENERGY	597
12	WORK AND POWER	608

## UNIT 5: SOLIDS, LIQUIDS AND GASES

13	DENSITY AND PRESSURE	621
14	SOLIDS, LIQUIDS AND GASES	629

## UNIT 6: MAGNETISM AND ELECTROMAGNETISM

15	MAGNETISM AND ELECTROMAGNETISM	639
16	ELECTRIC MOTORS AND ELECTROMAGNETIC INDUCTION	646

## UNIT 7: RADIOACTIVITY AND PARTICLES

17	ATOMS AND RADIOACTIVITY	657
18	RADIATION AND HALF-LIFE	669
19	APPLICATIONS OF RADIOACTIVITY	677
20	FISSION AND FUSION	686

## UNIT 8: ASTROPHYSICS

21	MOTION IN THE UNIVERSE	695
22	STELLAR EVOLUTION	701

# ABOUT THIS BOOK

This book is written for students following the Pearson Edexcel International GCSE (9–1) Science Double Award specification. You will need to study all of the content in this book for your examinations, except anything in Extension boxes.

In each unit of this book, there are concise explanations and worked examples, plus numerous exercises that will help you build up confidence. The book also describes the methods for carrying out all of the required practicals.

The language throughout this textbook is graded for speakers of English as an additional language (EAL), with advanced Science-specific terminology highlighted and defined in the glossaries on the eBook. A list of command words at the back of the book will help you to learn the language you will need in your examinations.

You will also find that questions in this book have Progression icons and Skills tags. The Progression icons refer to Pearson's Progression scale. This scale – from 1 to 12 – tells you what level you have reached in your learning and will help you to see what you need to do to progress to the next level. Furthermore, Edexcel have developed a Skills grid showing the skills you will practise throughout your time on the course. The skills in the grid have been matched to questions in this book to help you see which skills you are developing. You can find Pearson's Progression scale at [www.pearsonglobalschools.com/igscienceprogression](http://www.pearsonglobalschools.com/igscienceprogression) along with guidelines on how to use it. Edexcel's skills grid can be found on their website.

*Learning Objectives* show what you will learn in each chapter.

*Unit boxes* tell you which units – for example, metres, grams and seconds – you will need to know and use for the study of a topic.

WAVES
PROPERTIES OF WAVES
599

### 7 PROPERTIES OF WAVES

Talking to someone using a mobile phone is something most of us do several times a day. The technology that had to be developed for this to happen was based on a thorough understanding of the properties of waves.

In this chapter you will learn about different types of waves and their properties (characteristics).



▲ Figure 7.1 Using microwaves to communicate

#### LEARNING OBJECTIVES

- Explain the difference between longitudinal and transverse waves
- Know the definitions of amplitude, wavefront, frequency, wavelength and period of a wave
- Know that waves transfer energy and information without transferring matter
- Know and use the relationship between the speed, frequency and wavelength of a wave:  
wave speed = frequency × wavelength  
 $v = f \times \lambda$
- Use the relationship between frequency and time period:  
frequency =  $\frac{1}{\text{time period}}$   
 $f = \frac{1}{T}$
- Use the above relationships in different contexts including sound waves and electromagnetic waves
- Explain that all waves can be reflected and refracted
- Explain why there is a change in the observed frequency and wavelength of a wave when its source is moving relative to an observer, and that this is known as the Doppler effect

#### UNITS

In this unit, you will need to use degrees (°) as the unit of angle, hertz (Hz) as the unit of frequency, metre (m) as the unit of length, metre per second (m/s) as the unit of speed and second (s) as the unit of time.

#### WHAT ARE WAVES?

Waves are a way of transferring energy from place to place. As we can see in Figure 7.1 we often use them to transfer information. All these transfers take place with no matter being transferred.



▲ Figure 7.2 Waves are produced if we drop a stone into a pond. The circular wavefronts spread out from the point of impact, carrying energy in all directions, but the water in the pond does not move from the centre to the edges.

*Looking Ahead* tells you what you would learn if you continued your study of Science to a higher level, such as International A Level.

420 PHYSICAL CHEMISTRY      ENERGETICS

**KEY POINT**  
You may sometimes see these plotted as smooth graphs, but that is technically wrong. A smooth curve should only be used for a continuous independent variable, one which can take any value. There is no such thing as an alcohol with 0.5 or 1.64 carbon atoms! The number of carbon atoms is a **non-continuous variable** because it can only take whole number values.

Figure 15.8 Molar enthalpy change of the combustion of alcohols.

**WORKING OUT ENTHALPY CHANGES FOR REACTIONS INVOLVING SOLUTIONS USING CALORIMETRY EXPERIMENTS**

You can use very similar methods for measuring molar enthalpy changes in displacement reactions (e.g. zinc and copper(II) sulfate solution), dissolving (e.g. dissolving ammonium chloride in water to form a solution) and neutralisation reactions (e.g. between potassium hydroxide solution and dilute hydrochloric acid). These experiments also involve heating water, but this time the water is part of the solutions we are using. We will look at how to do this in the next few practicals.

**ACTIVITY 2**

**PRACTICAL: MEASURING ENTHALPY CHANGES FOR DISPLACEMENT REACTIONS**

In order to determine the enthalpy change of the reaction of zinc and copper(II) sulfate, the following procedure could be used:

- Place a polystyrene cup in a 250 cm<sup>3</sup> glass beaker.
- Transfer 50 cm<sup>3</sup> of 0.200 mol/dm<sup>3</sup> copper(II) sulfate solution into the polystyrene cup using a measuring cylinder.
- Weigh 1.20g of zinc using a weighing boat on a balance.
- Record the initial temperature of the copper(II) sulfate solution.
- Add the zinc.
- Stir the solution as quickly as possible.
- Record the maximum temperature reached.

15.9 A calorimetry experiment to measure the enthalpy change of a displacement reaction.

**Safety Note:** Avoid skin contact with the chemicals.

Practicals describe the methods for carrying out all of the practicals you will need to know for your examination.

PHYSICAL CHEMISTRY      ENERGETICS      421

**SAMPLE DATA**

initial temperature of copper(II) sulfate solution / °C	17.0
maximum temperature of the reaction mixture / °C	27.3

We can use this data to calculate the enthalpy change for this displacement reaction, when 1 mole of copper(II) sulfate reacts with zinc.

Heat given out in this reaction:  $Q = mc\Delta T = 50 \times 4.18 \times (27.3 - 17.0)$   
 $= 2157.7 \text{ J}$   
 $= 2.15277 \text{ kJ}$

Here we assume the following:

- The density of the copper sulfate solution is the same as that of water, so 1 cm<sup>3</sup> of solution has a mass of 1 g.
- The specific heat capacity of the mixture is the same as that of water. This is a fairly reasonable assumption because the reaction mixture is mostly water.

In this experiment we have used excess zinc. Excess means more than enough zinc is present to ensure all the copper(II) sulfate reacts. If you calculate the number of moles of copper(II) sulfate and the number of moles of zinc used in this procedure, you should spot that the number of moles of zinc used is more than that of copper(II) sulfate:

$$\text{number of moles (n)} \text{ of zinc added} = \frac{\text{mass (m)}}{\text{relative atomic mass (A)}}$$

$$= \frac{1.20}{65}$$

$$= 0.0185 \text{ mol}$$

number of moles (n) of copper(II) sulfate added = volume (V) × concentration (C)  
 $= 0.050 \times 0.200$   
 $= 0.0100 \text{ mol}$

Now we need to calculate how much heat is released when 1 mole of copper sulfate reacts with excess zinc:

Molar enthalpy change of reaction ( $\Delta H$ )

$$\Delta H = \frac{\text{heat energy change (Q)}}{\text{number of moles of copper sulfate reacted (n)}}$$

$$= \frac{2.1527}{0.0100} = 215.27 \text{ kJ/mol}$$

The amount of heat released in the displacement reaction when 1 mole of CuSO<sub>4</sub> reacts with excess Zn is therefore:

$$\text{Zn(s)} + \text{CuSO}_4(\text{aq}) \rightarrow \text{ZnSO}_4(\text{aq}) + \text{Cu(s)} \quad \Delta H = -215 \text{ kJ/mol}$$

We have added the negative sign because we know that the temperature of the reaction mixture went up. The negative sign shows that this is an exothermic reaction; heat is released.

**KEY POINT**  
1 mol of Zn reacts with 1 mol of CuSO<sub>4</sub>, therefore 0.0100 mol of CuSO<sub>4</sub> reacts with 0.0100 mol of Zn. We added 0.0185 mol of Zn, so Zn is in excess.

**REMINISCE**  
Remember that we have to divide the volume by 1000 to convert it to dm<sup>3</sup> because the concentration is given in mol/dm<sup>3</sup>.

**EXTENSION WORK**  
You could repeat this experiment with metals of different reactivities. The more reactive a metal is, the more heat should be released in the displacement reaction. Make sure you keep everything else the same: the number of moles of the metals, the size of the solid particles, and the volume and concentration of the copper(II) sulfate solution. Do not use metals that are more reactive than magnesium, otherwise you are measuring the heat released when the metal reacts with water instead!

Extension work boxes include content that is not on the specification and which you do not have to learn for your examination. However, it will help to extend your understanding of the topic.

158 PLANT PHYSIOLOGY      REPRODUCTION IN PLANTS

**SKILLS REASONING**

7 The diagram shows a potato plant producing new tubers (potatoes). Buds on the parent plant grow into stems that grow downwards, called stolons. The ends of each stolon develop into a new tuber.

- Give two pieces of evidence which show that this is an asexual method of reproduction.
- Explain why all the new tubers will be genetically identical.
- Even though the tubers are genetically identical, the plants that grow from them may not be the same height. Explain why.
- Why do wild plants need to reproduce sexually as well as asexually?

**SKILLS CRITICAL THINKING**

8 The drawing shows a wind-pollinated flower.

- Name the structures labelled A, B, C and D.
- Give three pieces of evidence visible in the diagram, which show that this flower is wind-pollinated.
- Describe how fertilisation takes place once a flower has been pollinated.
- Describe four ways in which you would expect an insect-pollinated flower to be different from the flower shown.

**SKILLS REASONING**

9 The drawing shows a strawberry plant reproducing in two ways.

- Which of the two methods of reproduction shown will result in offspring that show genetic variation? Explain your answer.
- Is the strawberry flower likely to be wind-pollinated or insect-pollinated? Give reasons for your answer.

10 All banana plants are reproduced asexually. Biologists are concerned for their future, as a new strain of fungus has appeared which is killing all the banana plants in some plantations.

- Explain why the fungus is able to kill all the banana plants in some plantations.
- Explain why this would be less likely to happen if banana plants reproduced sexually.
- Describe the benefits of reproducing banana plants asexually.

Skills tags tell you which skills you are practising in each question.

Chapter Questions test your knowledge of the topic in that chapter.

Progression icons show the level of difficulty according to the Pearson International GCSE Science Progression Scale.

Unit Questions test your knowledge of the whole unit and provide quick, effective feedback on your progress.

PLANT PHYSIOLOGY      UNIT QUESTIONS      159

**UNIT QUESTIONS**

**SKILLS ANALYSIS**

1 Light intensity and the concentration of carbon dioxide in the atmosphere influence the rate of photosynthesis.

The graph shows the effect of changing light intensity on the rate of photosynthesis at two different carbon dioxide concentrations.

- Describe the effect of light intensity on the rate of photosynthesis at each concentration of carbon dioxide up to light intensity X and beyond light intensity X. (4)
- Which factor limits the rate of photosynthesis up to light intensity X and beyond light intensity X? (2)
- Explain your answer in each case. (2)
- Describe two other factors which influence the rate of photosynthesis. (2)
- Explain why each is a limiting factor. (4)
- 'Photosynthesis is a means of converting light energy into chemical energy.' Explain what this statement means. (2)

**(Total 17 marks)**

**SKILLS CRITICAL THINKING**

**SKILLS INTERPRETATION**

**SKILLS DECISION MAKING**

2 In an investigation to determine the water potential of potato cells, the following procedure was adopted.

Cylinders of tissue were obtained from a potato. Each was the same diameter and cut to a length of 5 cm.

Each cylinder was dried and then weighed.

Three potato cylinders were placed in each of six different concentrations of sucrose solution and left for two hours.

The cylinders were then removed from the solutions, dried and reweighed. The percentage change in mass for each was calculated, and then an average percentage change in mass calculated for each solution.

The graph summarises the results.

# ASSESSMENT OVERVIEW

The following tables give an overview of the assessment for this course.

We recommend that you study this information closely to help ensure that you are fully prepared for this course and know exactly what to expect in the assessment.

BIOLOGY PAPER 1	SPECIFICATION	PERCENTAGE	MARK	TIME	AVAILABILITY
Written examination paper Paper code 4SD0/1B Externally set and assessed by Edexcel	Science Double Award	33.3%	110	2 hours	January and June examination series First assessment June 2019
CHEMISTRY PAPER 1	SPECIFICATION	PERCENTAGE	MARK	TIME	AVAILABILITY
Written examination paper Paper code 4SD0/1C Externally set and assessed by Edexcel	Science Double Award	33.3%	110	2 hours	January and June examination series First assessment June 2019
PHYSICS PAPER 1	SPECIFICATION	PERCENTAGE	MARK	TIME	AVAILABILITY
Written examination paper Paper code 4SD0/1P Externally set and assessed by Edexcel	Science Double Award	33.3%	110	2 hours	January and June examination series First assessment June 2019

## ASSESSMENT OBJECTIVES AND WEIGHTINGS

ASSESSMENT OBJECTIVE	DESCRIPTION	% IN INTERNATIONAL GCSE
AO1	Knowledge and understanding of science	38%–42%
AO2	Application of knowledge and understanding, analysis and evaluation of science	38%–42%
AO3	Experimental skills, analysis and evaluation of data and methods in science	19%–21%

## EXPERIMENTAL SKILLS

In the assessment of experimental skills, students may be tested on their ability to:

- solve problems set in a practical context
- apply scientific knowledge and understanding in questions with a practical context
- devise and plan investigations, using scientific knowledge and understanding when selecting appropriate techniques
- demonstrate or describe appropriate experimental and investigative methods, including safe and skilful practical techniques
- make observations and measurements with appropriate precision, record these methodically and present them in appropriate ways
- identify independent, dependent and control variables
- use scientific knowledge and understanding to analyse and interpret data to draw conclusions from experimental activities that are consistent with the evidence
- communicate the findings from experimental activities, using appropriate technical language, relevant calculations and graphs
- assess the reliability of an experimental activity
- evaluate data and methods taking into account factors that affect accuracy and validity.

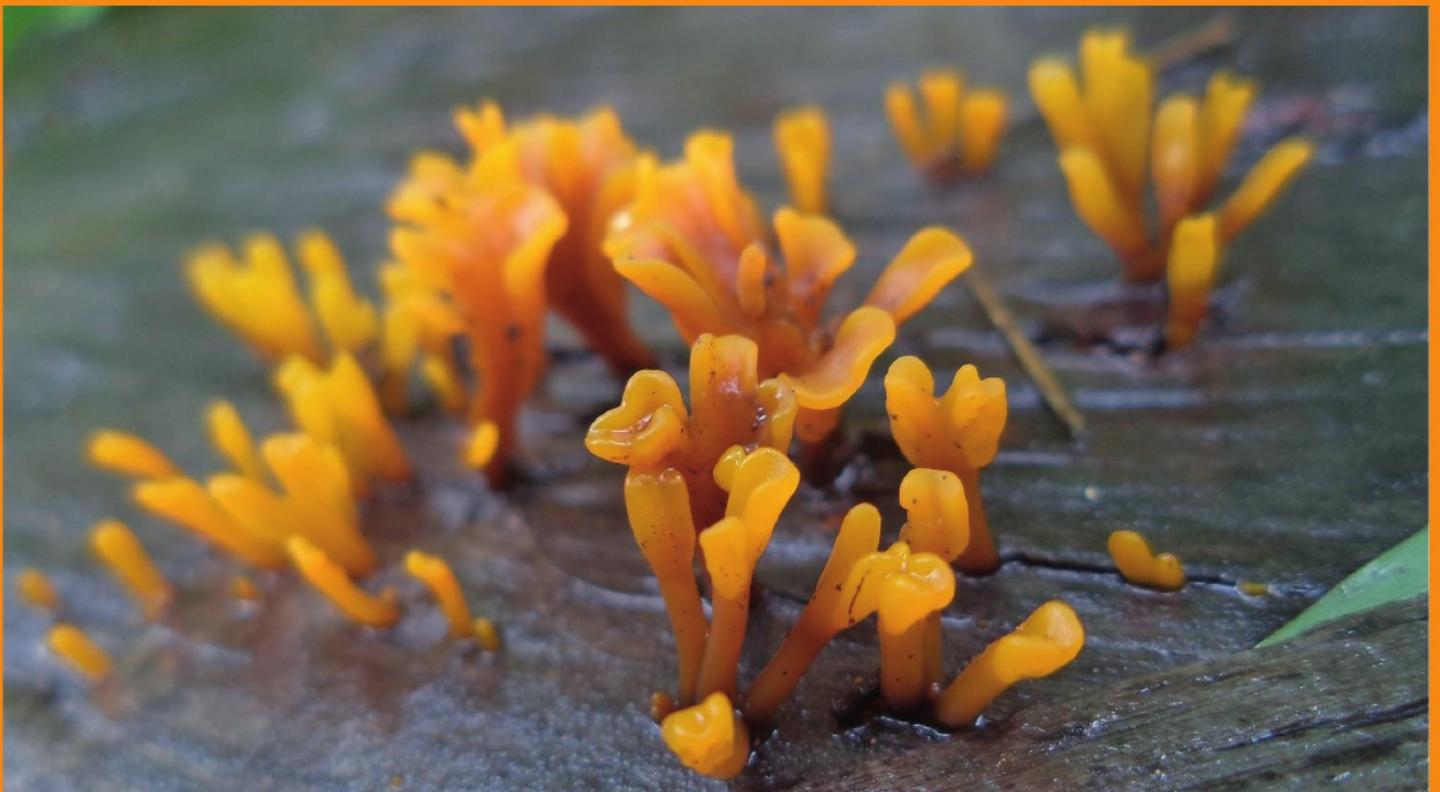
## CALCULATORS

Students are permitted to take a suitable calculator into the examinations. Calculators with QWERTY keyboards or that can retrieve text or formulae will not be permitted.

# BIOLOGY UNIT 1

# ORGANISMS AND LIFE PROCESSES

All living organisms are composed of microscopic units known as cells. These building blocks of life have a number of features in common, which allow them to grow, reproduce, and generate more organisms. In Chapter 1 we start by looking at the structure and function of cells, and the essential life processes that go on within them. Despite the fact that cells are similar in structure, there are many millions of different species of organisms. Chapter 2 looks at the diversity of living things and how we can classify them into groups on the basis of the features that they show.



# 1 LIFE PROCESSES

There are structural features that are common to the cells of all living organisms. In this chapter you will find out about these features and look at some of the processes that keep cells alive.

## LEARNING OBJECTIVES

- Understand the characteristics shared by living organisms
- Describe cell structures and their functions, including the nucleus, cytoplasm, cell membrane, cell wall, mitochondria, chloroplasts, ribosomes and vacuole
- Know the similarities and differences in the structures of plant and animal cells
- Understand the role of enzymes as biological catalysts in metabolic reactions
- Understand how temperature changes can affect enzyme function, including changes to the shape of the active site
- Understand how enzyme function can be affected by changes in pH altering the active site
- Investigate how enzyme activity can be affected by changes in temperature
- Describe the differences between aerobic and anaerobic respiration
- Understand how the process of respiration produces ATP in living organisms
- Know that ATP provides energy for cells
- Know the word equation and balanced chemical symbol equation for aerobic respiration
- Know the word equations for anaerobic respiration
- Investigate the evolution of carbon dioxide and heat from respiring seeds or other suitable living organisms
- Understand the processes of diffusion, osmosis and active transport by which substances move into and out of cells
- Understand how factors affect the rate of movement of substances into and out of cells
- Investigate diffusion in a non-living system (agar jelly)
- Describe the levels of organisation within organisms – organelles, cells, tissues, organ systems

All living organisms are composed of units called **cells**. The simplest organisms are made from single cells (Figure 1.1) but more complex plants and animals are composed of millions of cells. In many-celled (**multicellular**) organisms, there may be hundreds of different types of cells with different structures. They are specialised so that they can carry out particular functions in the animal or plant. Despite all the differences, there are basic features that are the same in all cells.



▲ Figure 1.1 Many simple organisms have 'bodies' made from single cells. Here are four examples.

There are eight life processes which take place in most living things.

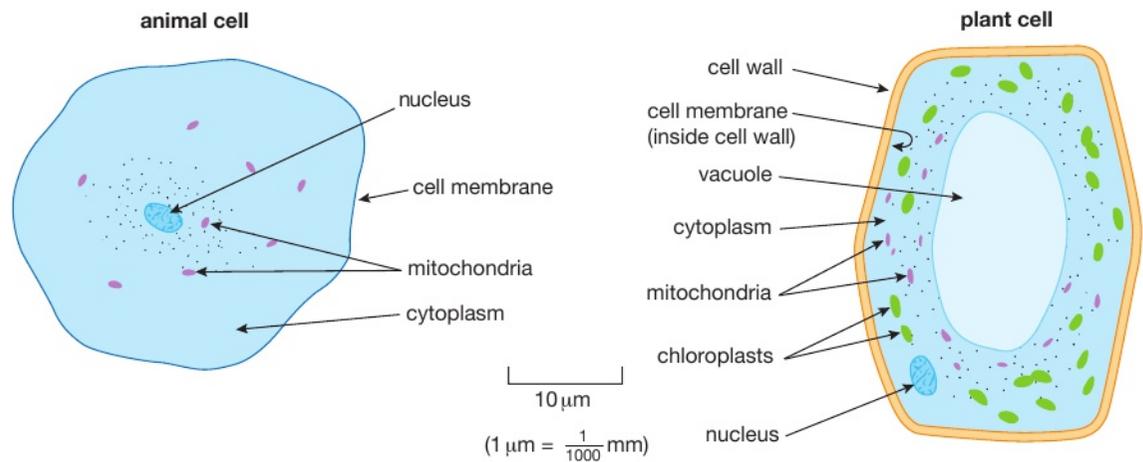
Organisms:

- require nutrition – plants make their own food, animals eat other organisms
- respire – release energy from their food
- excrete – get rid of waste products
- respond to stimuli – are sensitive to changes in their surroundings
- move – by the action of muscles in animals, and slow growth movements in plants
- control their internal conditions – maintain a steady state inside the body
- reproduce – produce offspring
- grow and develop – increase in size and complexity, using materials from their food.

## CELL STRUCTURE

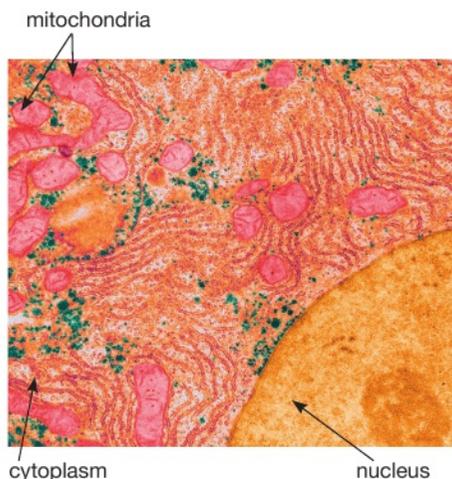
This part of the book describes the cell structure of 'higher' organisms such as animals, plants and fungi. The cells of bacteria are simpler in structure and will be described in Chapter 2.

Most cells contain certain parts such as the nucleus, cytoplasm and cell membrane. Some cells have structures missing, for instance red blood cells are unusual in that they have no nucleus. The first chapter in a biology textbook will usually present diagrams of 'typical' plant and animal cells. In fact, there is really no such thing as a 'typical' cell. Humans, for example, are composed of hundreds of different kinds of cells – from nerve cells to blood cells, skin cells to liver cells. What we really mean by a 'typical' cell is a general diagram that shows all the features that you will find in most cells (Figure 1.2). However, not all these are present in *all* cells – for example the cells in the parts of a plant that are not green do not contain chloroplasts.



▲ Figure 1.2 The structure of a 'typical' animal and plant cell.

The living material that makes up a cell is called **cytoplasm**. It has a texture rather like sloppy jelly, in other words somewhere between a solid and a liquid. Unlike a jelly, it is not made of one substance but is a complex material made of many different structures. You can't see many of these structures under an ordinary light microscope. An electron microscope has a much higher magnification and can show the details of these structures, which are called **organelles** (Figure 1.3).



▲ Figure 1.3 The organelles in a cell can be seen using an electron microscope.

The largest organelle in the cell is the **nucleus**. Nearly all cells have a nucleus. The few types that don't are usually dead (e.g. the xylem vessels in a stem, Chapter 11) or don't live for very long (e.g. red blood cells, Chapter 5). The nucleus controls the activities of the cell. It contains **chromosomes** (46 in human cells) which carry the genetic material, or **genes**. Genes control the activities in the cell by determining which proteins the cell can make. The DNA remains in the nucleus, but the instructions for making proteins are carried out of the nucleus to the cytoplasm, where the proteins are assembled on tiny structures called **ribosomes**. A cell contains thousands of ribosomes, but they are too small to be seen through a light microscope.

One very important group of proteins found in cells are **enzymes**. Enzymes control the chemical reactions that take place in the cytoplasm.

All cells are surrounded by a **cell membrane**, sometimes called the cell *surface* membrane to distinguish it from other membranes inside the cell. This is a thin layer like a 'skin' on the surface of the cell. It forms a boundary between the cytoplasm of the cell and the outside. However, it is not a complete barrier. Some chemicals can pass into the cell and others can pass out. We say that the membrane is **partially permeable**. The membrane can go further than this and actually *control* the movement of some substances – it is **selectively permeable**.

One organelle that is found in the cytoplasm of all living cells is the **mitochondrion** (plural mitochondria). In cells that need a lot of energy such as muscle or nerve cells, there are many mitochondria. This gives us a clue to their function. They carry out some of the reactions of **respiration** (see page 11) releasing energy that the cell can use. Most of the energy from respiration is released in the mitochondria.

## PLANT CELLS

All of the structures you have seen so far are found in both animal and plant cells. However, some structures are only ever found in plant cells. There are three in particular – the cell wall, a permanent vacuole and chloroplasts.

The **cell wall** is a layer of non-living material that is found outside the cell membrane of plant cells. It is made mainly of a carbohydrate called **cellulose**, although other chemicals may be added to the wall in some cells. Cellulose is a tough material that helps the cell keep its shape and is one reason why the 'body' of a plant has a fixed shape. Animal cells do not have a cell wall and tend to be more variable in shape. Plant cells absorb water, producing an internal pressure that pushes against adjacent cells, giving the plant support (see Chapter 11). Without a cell wall strong enough to resist these pressures, this method of support would be impossible. The cell wall is porous, so it is not a barrier to water or dissolved substances. We call it *freely permeable*.

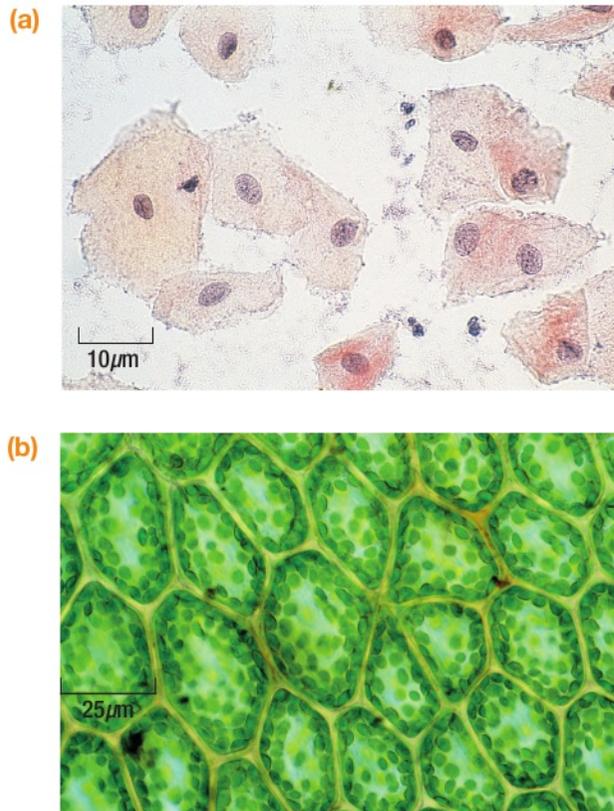
Mature (fully grown) plant cells often have a large central space surrounded by a membrane, called a **vacuole**. This vacuole is a permanent feature of the cell. It is filled with a watery liquid called cell sap, which is a store of dissolved sugars, mineral ions and other solutes. Animal cells do contain vacuoles, but they are only small, temporary structures.

Cells of the green parts of plants, especially the leaves, contain another very important organelle, the **chloroplast**. Chloroplasts absorb light energy to make food in the process of photosynthesis (see Chapter 10). They contain a green pigment called **chlorophyll**. Cells from the parts of a plant that are not green, such as the flowers, roots and woody stems, have no chloroplasts.

### KEY POINT

Nearly all cells contain cytoplasm, a nucleus, a cell membrane and mitochondria. As well as these structures, plant cells have a cell wall and a permanent vacuole, and plant cells that photosynthesise contain chloroplasts.

Figure 1.4 shows some animal and plant cells seen through the light microscope.



▲ Figure 1.4 (a) Cells from the lining of a human cheek. (b) Cells from the photosynthetic tissue of a leaf.

#### KEY POINT

The chemical reactions taking place in a cell are known as metabolic reactions. The sum of all the metabolic reactions is known as the **metabolism** of the cell. The function of enzymes is to catalyse metabolic reactions.

#### KEY POINT

You have probably heard of enzymes being involved in digestion of food. In the intestine enzymes are secreted onto the food to break it down. These are called *extracellular* enzymes, which means they function 'outside cells'. However, most enzymes stay inside cells and carry out their function there; they are *intracellular*. You will find out about digestive enzymes in Chapter 4.

#### KEY POINT

**Secretion** is the release of a fluid or substances from a cell or tissue.

## ENZYMES: CONTROLLING REACTIONS IN THE CELL

The chemical reactions that take place in a cell are controlled by a group of proteins called enzymes. Enzymes are biological **catalysts**. A catalyst is a chemical which speeds up a reaction without being used up itself. It takes part in the reaction, but afterwards is unchanged and free to catalyse more reactions. Cells contain hundreds of different enzymes, each catalysing a different reaction. This is how the activities of a cell are controlled – the nucleus contains the genes, which control the production of enzymes, which then catalyse reactions in the cytoplasm:

genes → proteins (enzymes) → catalyse reactions

Everything a cell does depends on which enzymes it can make, which in turn depends on which genes in its nucleus are working.

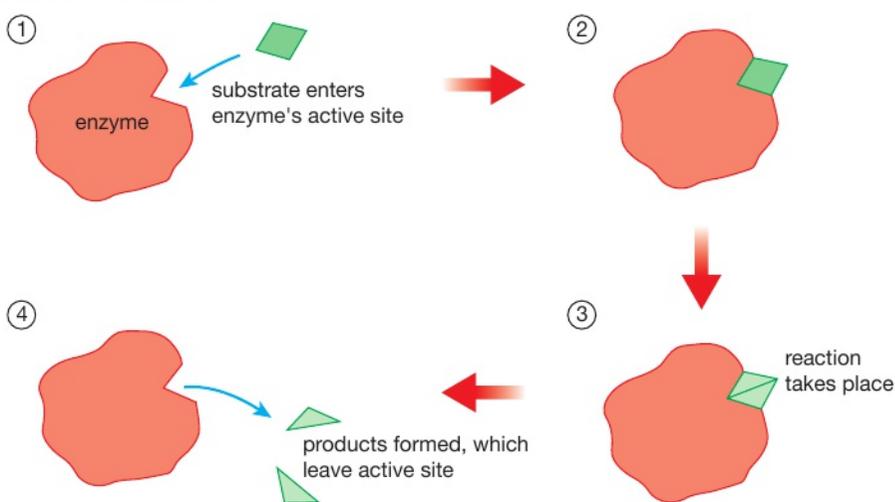
What hasn't been mentioned is why enzymes are needed at all. They are necessary because the temperatures inside organisms are low (e.g. the human body temperature is about 37 °C) and without catalysts, most of the reactions that happen in cells would be far too slow to allow life to go on. The reactions can only take place quickly enough when enzymes are present to speed them up.

It is possible for there to be thousands of different sorts of enzymes because they are proteins, and protein molecules have an enormous range of structures and shapes (see Chapter 4).

The molecule that an enzyme acts on is called its **substrate**. Each enzyme has a small area on its surface called the **active site**. The substrate attaches to the active site of the enzyme. The reaction then takes place and products are formed. When the substrate joins up with the active site it lowers the energy needed for the reaction to start, allowing the products to be formed more easily.

Enzymes also catalyse reactions where large molecules are built up from smaller ones. In this case, several substrate molecules attach to the active site, the reaction takes place and the larger product molecule is formed. The product then leaves the active site.

The substrate fits into the active site of the enzyme like a key fitting into a lock. Just as a key will only fit one lock, a substrate will only fit into the active site of a particular enzyme. This is known as the **lock and key model** of enzyme action. It is the reason why enzymes are *specific*, i.e. an enzyme will only catalyse one reaction.



▲ Figure 1.5 Enzymes catalyse reactions at their active site. This acts like a 'lock' to the substrate 'key'. The substrate fits into the active site, and products are formed. This happens more easily than without the enzyme – so enzymes act as catalysts.

After an enzyme molecule has catalysed a reaction, the product is released from the active site, and the enzyme is free to act on more substrate molecules.

## FACTORS AFFECTING ENZYMES

### KEY POINT

'Optimum' temperature means the 'best' temperature, in other words the temperature at which the reaction takes place most rapidly.

### DID YOU KNOW?

Kinetic energy is the energy an object has because of its movement. The molecules of enzyme and substrate are moving faster, so they have more kinetic energy.

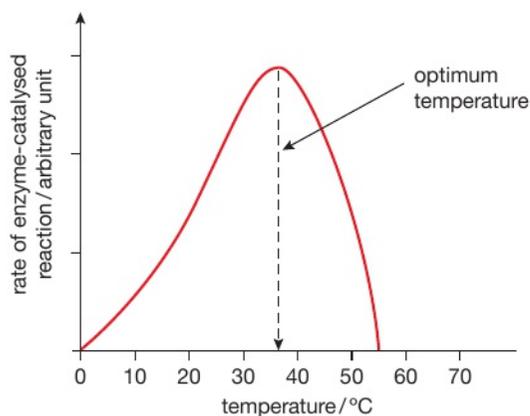
A number of factors affect the activity of enzymes. The rate of reaction may be increased by raising the concentration of the enzyme or the substrate. Two other factors that affect enzymes are temperature and pH.

### TEMPERATURE

The effect of temperature on the action of an enzyme is easiest to see as a graph, where we plot the rate of the reaction against temperature (Figure 1.6).

Enzymes in the human body have evolved to work best at body temperature (37 °C). The graph in Figure 1.6 shows a peak on the curve at this temperature, which is called the *optimum temperature* for the enzyme.

As the enzyme is heated up to the optimum temperature, the rise in temperature increases the rate of reaction. This is because higher temperatures give the molecules of the enzyme and the substrate more kinetic energy, so they collide more often. More collisions means that the reaction will take place more frequently.



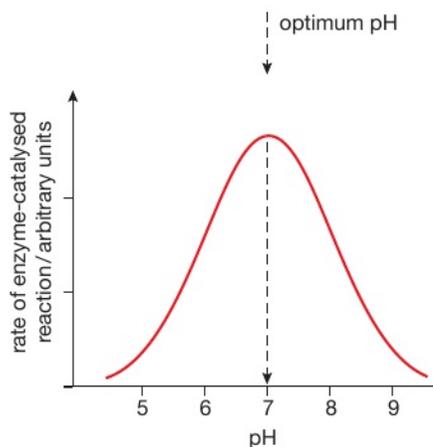
▲ Figure 1.6 Effect of temperature on the action of an enzyme.

However, above the optimum, temperature starts to have another effect. Enzymes are made of protein, and proteins are broken down by heat. From 40 °C upwards, the heat destroys the enzyme. We say that it is denatured. You can see the effect of **denaturing** when you boil an egg. The egg white is made of protein, and turns from a clear runny liquid into a white solid as the heat denatures the protein. Denaturing changes the shape of the active site so that the substrate will no longer fit into it. Denaturing is permanent – the enzyme molecules will no longer catalyse the reaction.

Not all enzymes have an optimum temperature near 37 °C, only those of animals such as mammals and birds, which all have body temperatures close to this value. Enzymes have evolved to work best at the normal body temperature of the organism. Bacteria that always live at an average temperature of 10 °C will probably have enzymes with an optimum temperature near 10 °C.

### pH

The pH around the enzyme is also important. The pH inside cells is neutral (pH 7) and most enzymes have evolved to work best at this pH. At extremes of pH either side of neutral, the enzyme activity decreases, as shown in Figure 1.7. The pH at which the enzyme works best is called its *optimum pH*. Either side of the optimum, the pH affects the structure of the enzyme molecule and changes the shape of its active site, so that the substrate will not fit into it so well.



▲ Figure 1.7 Most enzymes work best at a neutral pH.

#### KEY POINT

Although most enzymes work best at a neutral pH, a few have an optimum below or above pH 7. The stomach produces hydrochloric acid, which makes its contents very acidic (see Chapter 4). Most enzymes stop working at a low pH, but the stomach makes an enzyme called pepsin which has an optimum pH of about 2, so that it is adapted to work well in these unusually acidic surroundings.



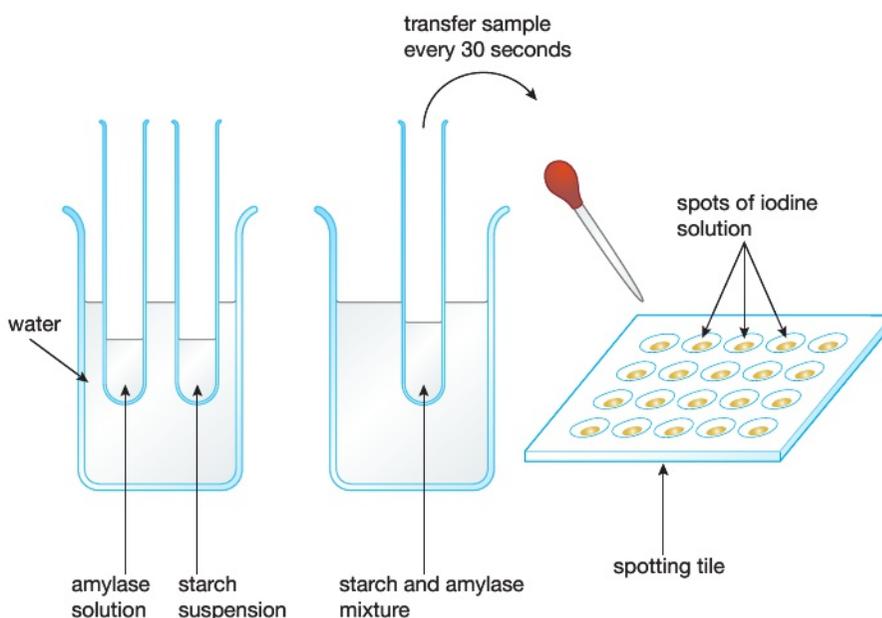
**Safety note:** Wear eye protection and avoid skin contact with the liquids. Amylase is hazardous to the eyes.

## ACTIVITY 1

### ▼ PRACTICAL: AN INVESTIGATION INTO THE EFFECT OF TEMPERATURE ON THE ACTIVITY OF AMYLASE

The digestive enzyme amylase breaks down starch into the sugar maltose. If the speed at which the starch disappears is recorded, this is a measure of the activity of the amylase.

Figure 1.8 shows apparatus which can be used to record how quickly the starch is used up.



▲ Figure 1.8 Investigating the breakdown of starch by amylase at different temperatures.

Spots of iodine solution are placed into the dips on the spotting tile. Using a syringe, 5 cm<sup>3</sup> of starch suspension is placed in one boiling tube, and 5 cm<sup>3</sup> of amylase solution in another tube, using a different syringe. The beaker is filled with water at 20 °C. Both boiling tubes are placed in the beaker of water for 5 minutes, and the temperature recorded.

The amylase solution is then poured into the starch suspension, leaving the tube containing the mixture in the water bath. Immediately, a small sample of the mixture is removed from the tube with a pipette and added to the first drop of iodine solution on the spotting tile. The colour of the iodine solution is recorded.

A sample of the mixture is taken every 30 seconds for 10 minutes and tested for starch as above, until the iodine solution remains yellow, showing that all the starch is used up.

The experiment is repeated, maintaining the water bath at different temperatures between 20 °C and 60 °C. A set of results is shown in the table below.

Time / min	Colour of mixture at different temperatures / (°C)				
	20	30	40	50	60
0.0	Blue-black	Blue-black	Blue-black	Blue-black	Blue-black
0.5	Blue-black	Blue-black	Brown	Blue-black	Blue-black
1.0	Blue-black	Blue-black	Yellow	Blue-black	Blue-black
1.5	Blue-black	Blue-black	Yellow	Blue-black	Blue-black
2.0	Blue-black	Blue-black	Yellow	Brown	Blue-black
2.5	Blue-black	Blue-black	Yellow	Brown	Blue-black
3.0	Blue-black	Blue-black	Yellow	Brown	Blue-black
3.5	Blue-black	Blue-black	Yellow	Yellow	Blue-black
4.0	Blue-black	Blue-black	Yellow	Yellow	Blue-black
5.5	Blue-black	Blue-black	Yellow	Yellow	Blue-black
6.0	Blue-black	Brown	Yellow	Yellow	Blue-black
6.5	Blue-black	Brown	Yellow	Yellow	Blue-black
7.0	Blue-black	Yellow	Yellow	Yellow	Blue-black
7.5	Blue-black	Yellow	Yellow	Yellow	Brown
8.0	Blue-black	Yellow	Yellow	Yellow	Brown
8.5	Brown	Yellow	Yellow	Yellow	Yellow
9.0	Brown	Yellow	Yellow	Yellow	Yellow
9.5	Yellow	Yellow	Yellow	Yellow	Yellow
10.0	Yellow	Yellow	Yellow	Yellow	Yellow

The rate of reaction can be calculated from the time taken for the starch to be fully broken down, as shown by the colour change from blue-black to yellow. For example, at 50 °C the starch had all been digested after 3.5 minutes. The rate is found by dividing the volume of the starch (5 cm<sup>3</sup>) by the time:

$$\text{Rate} = \frac{5.0 \text{ cm}^3}{3.5 \text{ min}} = 1.4 \text{ cm}^3 \text{ per min}$$

Plotting a graph of rate against temperature should produce a curve similar to the one shown in Figure 1.6. Try this, using the results in the table. Better still, you may be able to do this experiment and provide your own results.

If the curve doesn't turn out quite like the one in Figure 1.6, can you suggest why this is? How could you improve the experiment to get more reliable results?

## HOW THE CELL GETS ITS ENERGY

A cell needs a source of energy in order to be able to carry out all the processes needed for life. It gets this energy by breaking down food molecules to release the stored chemical energy that they contain. This process is called respiration. Many people think that respiration means the same as 'breathing', but although there are links between the two processes, the biological meaning of respiration is very different.

### KEY POINT

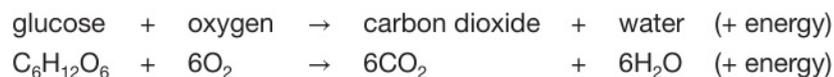
Respiration is called an *oxidation* reaction, because oxygen is used to break down food molecules.

Respiration happens in all the cells of our body. Oxygen is used to oxidise food, and carbon dioxide (and water) are released as waste products. The main food oxidised is a sugar called **glucose**. Glucose contains stored chemical energy that can be converted into other forms of energy that the cell can use. It is rather like burning a fuel to get the energy out of it, except that burning releases most of the energy as heat. Respiration releases some heat energy, but most is used to make a substance called ATP (see below). The energy stored in the ATP molecules can then be used for a variety of purposes, such as:

- contraction of muscle cells, producing movement
- active transport of molecules and ions (see page 17)
- building large molecules, such as proteins
- cell division.

The energy released as heat is also used to maintain a constant body temperature in mammals and birds (see Chapter 8).

The overall reaction for respiration is:



### DID YOU KNOW?

Carbon from the respired glucose passes out into the atmosphere as carbon dioxide. The carbon can be traced through this pathway using a radioactive form of carbon called carbon-14.

This is called **aerobic respiration**, because it uses oxygen. Aerobic respiration happens in the cells of humans and those of animals, plants and many other organisms. It is important to realise that the equation above is only a *summary* of the process. It actually takes place gradually, as a sequence of small steps, which release the energy of the glucose in small amounts. Each step in the process is catalysed by a different enzyme. The later steps in the process are the aerobic ones, and these release the most energy. They happen in the mitochondria of the cell.

### ATP – THE ENERGY 'CURRENCY' OF THE CELL

Respiration releases energy while other cell processes use it up. Cells have a way of passing the energy from respiration to the other processes that need it. They do this using a chemical called **adenosine triphosphate** or **ATP**. ATP is present in all living cells.

ATP is composed of an organic molecule called adenosine attached to three phosphate groups. In a cell, ATP can be broken down losing one phosphate group and forming adenosine diphosphate or ADP (Figure 1.9 (a)).

(a) When energy is needed ATP is broken down into ADP and phosphate (P):



(b) During respiration ATP is made from ADP and phosphate:



▲ Figure 1.9 ATP is the energy ‘currency’ of the cell.

When this reaction takes place, chemical energy is released and can be used to drive metabolic processes that need it.

During respiration the opposite happens – energy from the oxidation of glucose is used to drive the reverse reaction and a phosphate is added onto ADP (Figure 1.9 (b)).

ATP is often described as the energy ‘currency’ of the cell. It transfers energy between the process that releases it (respiration) and the processes in a cell that use it up.

## ANAEROBIC RESPIRATION

There are some situations where cells can respire *without* using oxygen. This is called **anaerobic respiration**. In anaerobic respiration, glucose is not completely broken down, so less energy is released. The advantage of anaerobic respiration is that it can occur in situations where oxygen is in short supply. Two important examples of this are in yeast cells and muscle cells.

Yeasts are single-celled fungi. They are used in processes such as baking bread. When yeast cells are prevented from getting enough oxygen, they stop respiring aerobically and start to respire anaerobically instead. The glucose is partly broken down into ethanol (alcohol) and carbon dioxide:



This process is described in more detail in Chapter 21.

Think about the properties of ethanol – it makes a good fuel and will burn to produce a lot of heat, so it still has a lot of chemical energy ‘stored’ in it.

Muscle cells can also respire anaerobically when they are short of oxygen. If muscles are overworked, the blood cannot reach them fast enough to deliver enough oxygen for aerobic respiration. This happens when a person does a ‘burst’ of activity, such as a sprint, or quickly lifting a heavy weight. This time the glucose is broken down into a substance called **lactate**:



Anaerobic respiration provides enough energy to keep the overworked muscles going for a short period. During the exercise, the level of lactate rises in the muscle cells and bloodstream.

**DID YOU KNOW?**

It was once thought that lactate was toxic and caused muscle fatigue. We now know that this is *not* true. In fact physiologists have shown that lactate actually *delays* muscle fatigue. Fatigue is caused by other changes that happen in the muscles during exercise.

**DID YOU KNOW?**

Lactate is sometimes called lactic acid.

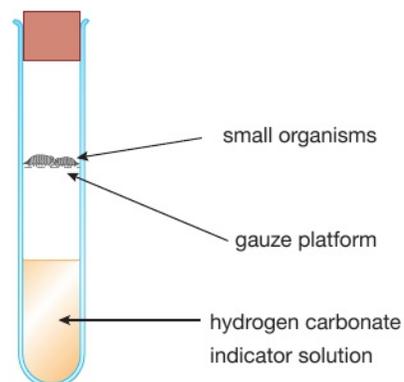
After the exercise the lactate is respired aerobically in the mitochondria. The volume of oxygen needed to completely oxidise the lactate that builds up in the body during anaerobic respiration is called the **oxygen debt**.

**ACTIVITY 2**

### ▼ PRACTICAL: DEMONSTRATION OF THE PRODUCTION OF CARBON DIOXIDE BY SMALL LIVING ORGANISMS

Hydrogen carbonate indicator solution is normally orange, but turns yellow if carbon dioxide is added to it. The indicator is sensitive to small changes in carbon dioxide concentration, and can be used to show production of carbon dioxide by small organisms such as woodlice, maggots (fly larvae) or germinating seeds.

The organisms are placed in a stoppered boiling tube with the indicator, as shown in Figure 1.10. The gauze platform supports the organisms above the hydrogen carbonate indicator solution and stops them from coming into contact with the chemical.



▲ Figure 1.10 Testing for carbon dioxide production by small organisms.

Of the three species of organisms mentioned above, which do you think would change the colour of the indicator most quickly? If you are able to observe each of the organisms, this might help with your prediction.

When you have made your prediction (called a 'hypothesis'), plan an investigation to test it. Take care to consider the variables that need to be **controlled variable**, and don't forget to include a description of a fourth tube that you would need to set up as the experimental Control (see Appendix A for an explanation of these terms).

It may be possible for you to carry out the investigation using similar apparatus and organisms.



**Safety note:** Wear eye protection, treat all living organisms with care and avoid skin contact.



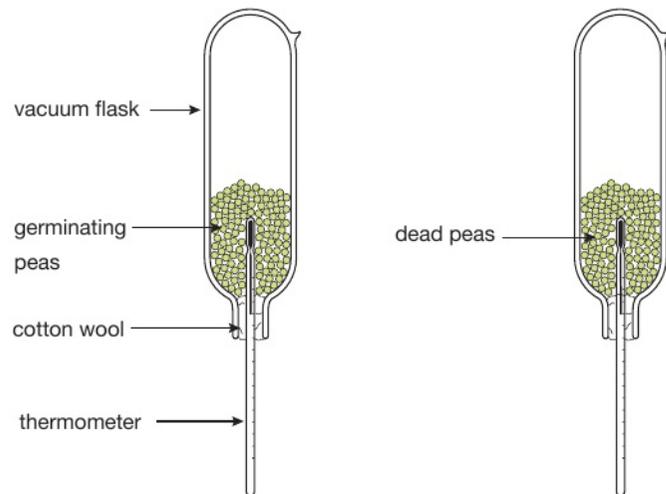
**Safety note:** Eye protection should be worn when using bleach to wash the peas and skin contact avoided.

### ACTIVITY 3

#### ▼ PRACTICAL: DEMONSTRATION THAT HEAT IS PRODUCED BY RESPIRATION

Some peas are soaked in water for 24 hours, so that they start to germinate. A second batch of peas is boiled, to kill them. Each set of peas is washed in a 1% bleach solution, which acts as a disinfectant, killing any bacteria present on the surface of the peas. The peas are then rinsed twice in distilled water to remove any traces of bleach.

Each batch of peas is placed in an inverted vacuum flask as shown in Figure 1.11, leaving some air in each flask. A vacuum flask insulates its contents, so that any small temperature change inside the flask can be measured.



▲ Figure 1.11 Experiment to show that heat is produced during respiration in germinating peas.

The seeds produce carbon dioxide gas, which is denser than air. The inverted flasks and cotton wool allow this to escape. It might otherwise kill the peas.

The apparatus is left set up for a couple of days, and the temperature inside each flask measured at the start and end of the experiment.

The following results were obtained from this experiment:

Temperature in both flasks at the start = 21 °C

Temperature in flask with dead peas at the end = 21 °C

Temperature in flask with living peas at the end = 24 °C

Can you explain these results? Why is it necessary to kill any microorganisms on the surface of the peas? Explain the importance of the flask containing dead peas.

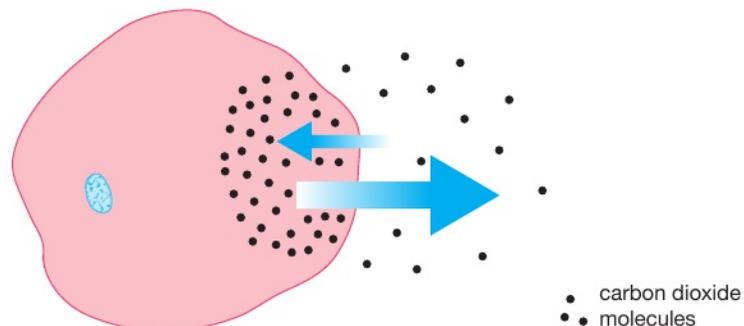
## MOVEMENT OF MATERIALS IN AND OUT OF CELLS

Cell respiration shows the need for cells to be able to take in certain substances from their surroundings, such as glucose and oxygen, and get rid of others, such as carbon dioxide and water. As you have seen, the cell surface membrane can control which chemicals can pass in and out – it is described as selectively permeable.

There are three main ways that molecules and ions can move through the membrane. They are diffusion, active transport and osmosis.

### DIFFUSION

Many substances can pass through the membrane by **diffusion**. Diffusion happens when a substance is more concentrated in one place than another. For example, if the cell is making carbon dioxide by respiration, the concentration of carbon dioxide inside the cell will be higher than outside. This difference in concentration is called a concentration gradient. The molecules of carbon dioxide are constantly moving about because of their kinetic energy. The cell membrane is permeable to carbon dioxide, so the molecules can move in either direction through it. Since there is a higher concentration of carbon dioxide molecules inside the cell than outside, over time, more molecules will move from inside to outside than move in the other direction. We say that there is a *net* movement of the molecules out of the cell (Figure 1.12).



▲ Figure 1.12 Carbon dioxide is produced by respiration, so its concentration builds up inside the cell. Although the carbon dioxide molecules diffuse in both directions across the cell membrane, the overall (net) movement is out of the cell, down the concentration gradient.

#### KEY POINT

Diffusion is the net movement of particles (molecules or ions) from a region of high concentration to a region of low concentration, i.e. down a concentration gradient.

The opposite happens with oxygen. Respiration uses up oxygen, so there is a concentration gradient of oxygen from outside to inside the cell. There is therefore a net movement of oxygen *into* the cell by diffusion.

The rate of diffusion is affected by various factors.

- The concentration gradient. Diffusion happens more quickly when there is a steep concentration gradient (i.e. a big difference in concentrations between two areas).
- The surface area to volume ratio. A larger surface area in proportion to the volume will increase the rate.
- The distance. The rate is decreased if the distance over which diffusion has to take place is greater.
- The temperature. The rate is greater at higher temperatures. This is because a high temperature provides the particles with more kinetic energy.



**Safety note:** Wear eye protection and avoid all skin contact with the acid and the dyed agar blocks.

## ACTIVITY 4

### ▼ PRACTICAL: DEMONSTRATION OF DIFFUSION IN A JELLY

**Agar** is a jelly that is used for growing cultures of bacteria. It has a consistency similar to the cytoplasm of a cell. Like cytoplasm, it has a high water content. Agar can be used to show how substances diffuse through a cell.

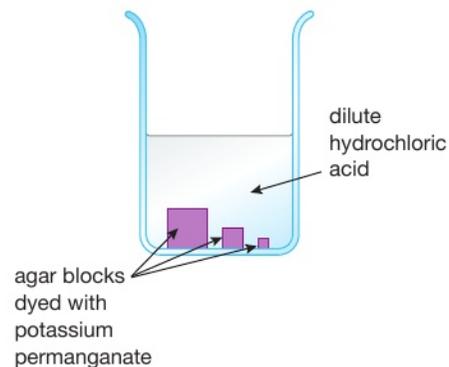
This demonstration uses the reaction between hydrochloric acid and potassium permanganate solution. When hydrochloric acid comes into contact with potassium permanganate, the purple colour of the permanganate disappears.

A Petri dish is prepared which contains a 2 cm deep layer of agar jelly, dyed purple with potassium permanganate. Three cubes of different sizes are cut out of the jelly, with side lengths 2 cm, 1 cm and 0.5 cm. The cubes have different volumes and total surface areas. They also have a different surface area to volume ratio, as shown in the table below.

Length of side of cube / cm	Volume of cube / cm <sup>3</sup> (length × width × height)	Surface area of cube / cm <sup>2</sup> (length × width of one side) × 6	Ratio of surface area to volume of cube (surface area divided by volume)
2	$(2 \times 2 \times 2) = 8$	$(2 \times 2) \times 6 = 24$	$24/8 = 3$
1	$(1 \times 1 \times 1) = 1$	$(1 \times 1) \times 6 = 6$	$6/1 = 6$
0.5	$(0.5 \times 0.5 \times 0.5) = 0.125$	$(0.5 \times 0.5) \times 6 = 1.5$	$1.5/0.125 = 12$

Notice that the smallest cube has the largest surface area to volume ratio. The same is true of cells – a small cell has a larger surface area to volume ratio than a large cell.

The cubes are carefully dropped, at the same time, into a beaker of dilute hydrochloric acid (Figure 1.13).



► Figure 1.13 Investigating diffusion in a jelly.

The time taken for each cube to turn colourless is noted.

Which cube would be the first to turn colourless and which the last? Explain the reasoning behind your prediction.

If the three cubes represented cells of different sizes, which cell would have the most difficulty in obtaining substances by diffusion?

It may be possible for you to try this experiment, using similar apparatus.

## ACTIVE TRANSPORT

Diffusion happens because of the kinetic energy of the particles. It does not need an 'extra' source of energy from respiration. However, sometimes a cell needs to take in a substance when there is very little of that substance outside the cell, in other words *against* a concentration gradient. It can do this by another process, called **active transport**.

During active transport a cell uses energy from respiration to take up substances, rather like a pump uses energy to move a liquid from one place to another. In fact, biologists speak of the cell 'pumping' ions or molecules in or out. The pumps are large protein molecules located in the cell membrane, and they are driven by the breakdown of ATP. An example of a place where this happens is in the human small intestine, where some glucose in the gut is absorbed into the cells lining the intestine by active transport. The roots of plants also take up certain mineral ions in this way. Cells use active transport to control the uptake of many substances.

## KEY POINT

Active transport is the movement of substances against a concentration gradient, using energy from respiration.

## OSMOSIS

Water moves across cell membranes by a special sort of diffusion, called **osmosis**. Osmosis happens when the total concentrations of all dissolved substances inside and outside the cell are different. Water will move across the membrane from the more dilute solution to the more concentrated one. Notice that this is still obeying the rules of diffusion – the water moves from where there is a higher concentration of *water* molecules to a lower concentration of *water* molecules. Osmosis can only happen if the membrane is permeable to water but not to some other solutes. We say that it is partially permeable.

Osmosis is important for moving water from cell to cell, for example in plant roots. You will find out more about osmosis in Chapter 11.

## KEY POINT

Osmosis in cells is the net movement of water from a dilute solution to a more concentrated solution across the partially permeable cell membrane.

## SPECIALISED EXCHANGE SURFACES

All cells exchange substances with their surroundings, but some parts of animals or plants are specially adapted for the exchange of materials because they have a very large surface area in proportion to their volume. In animals, two examples are the alveoli (air sacs) of the lungs (Chapter 3) and the villi of the small intestine (Chapter 4). Diffusion is a slow process, and organs that rely on diffusion need a large surface over which it can take place. The alveoli allow the exchange of oxygen and carbon dioxide to take place between the air and the blood during breathing. The villi of the small intestine provide a large surface area for the absorption of digested food. In plants, exchange surfaces are also adapted by having a large surface area, such as the spongy mesophyll of the leaf (Chapter 10) and the **root hairs**.

## KEY POINT

'Adapted' or 'an **adaptation**' means that the structure of a cell or an organism is suited to its function. It is a word that is very commonly used in biology, and will appear again in many of the chapters of this book. We also use it when we say that organisms are adapted to their environment.

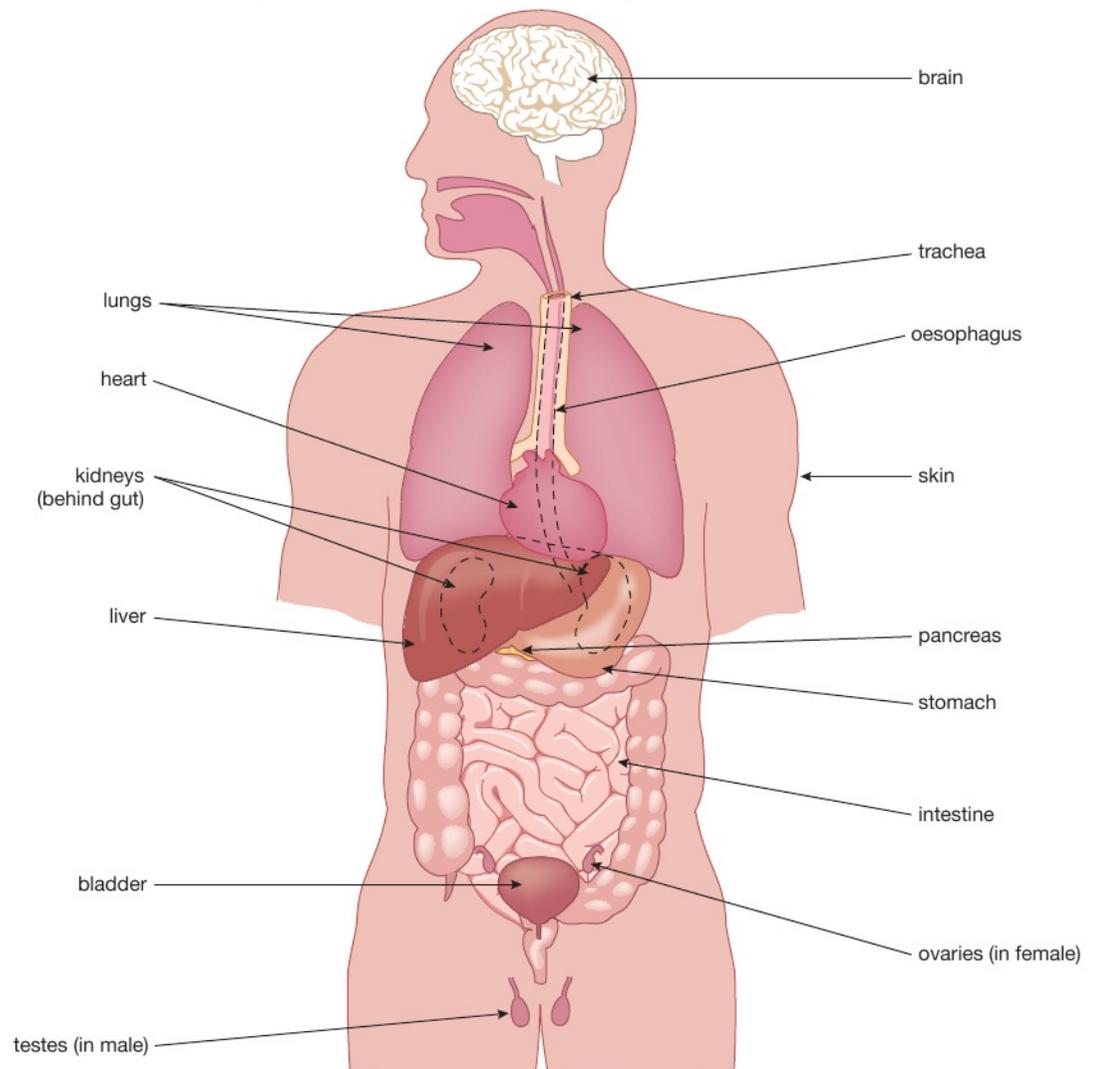
## CELLS, TISSUES AND ORGANS

Cells with a similar function are grouped together as **tissues**. For example, the muscle of your arm contains millions of similar muscle cells, all specialised for one function – contraction to move the arm bones. This is muscle tissue. However, a muscle also contains other tissues, such as blood, nervous tissue and epithelium (lining tissue). A collection of several tissues carrying out a particular function is called an **organ**. The main organs of the human body are shown in Figure 1.14. Plants also have tissues and organs. Leaves, roots, stems and flowers are all plant organs.

In animals, jobs are usually carried out by several different organs working together. This is called an **organ system**. For example, the digestive system consists of the gut, along with glands such as the pancreas and gall bladder.

The function of the whole system is to digest food and absorb the digested products into the blood. There are seven main systems in the human body. These are the:

- digestive system
- gas exchange system – including the lungs, which exchange oxygen and carbon dioxide
- circulatory system – including the heart and blood vessels, which transport materials around the body
- excretory system – including the kidneys, which filter toxic waste materials from the blood
- nervous system – consisting of the brain, spinal cord and nerves, which coordinate the body's actions
- endocrine system – glands secreting hormones, which act as chemical messengers
- reproductive system – producing sperm in males and eggs in females, and allowing the development of the embryo.



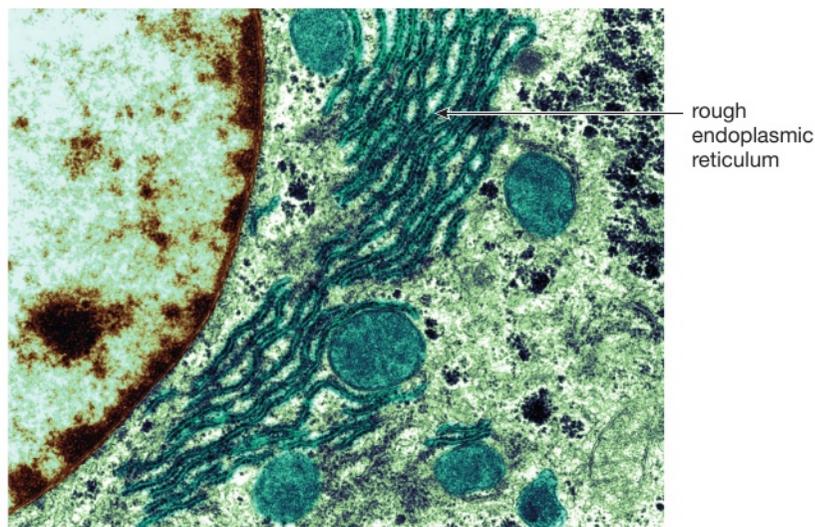
▲ Figure 1.14 Some of the main organs of the human body.

## LOOKING AHEAD – MEMBRANES IN CELLS

If you continue to study biology beyond International GCSE, you will learn more about the structure and function of cells. You might like to look on the Internet for some electron micrographs and carry out some further research into cells.

Electron micrographs allow us to see cells at a much greater magnification than by using a light microscope. They also reveal more detail. The image produced by a light microscope can only distinguish features about the size of a mitochondrion. The electron microscope has a much greater *resolution*. Resolution is the ability to distinguish two points in an image as being separate. The maximum resolution of a light microscope is about 200 nanometres (nm), while with an electron microscope we can distinguish structures less than 1 nm in size. That is why ribosomes are only visible using an electron microscope – they are about 25 nm in diameter. A nanometre (nm) is  $10^{-9}$  m, or one millionth of a millimetre.

Electron microscopy reveals that much of the cytoplasm is made up of membranes. As well as the cell surface membrane, there are membranes around organelles such as the nucleus, mitochondria and chloroplasts. In addition, there is an extensive system of membranes running throughout the cytoplasm, called the endoplasmic reticulum (ER). Some ER is covered in ribosomes, and is called rough ER (Figure 1.15).



▲ Figure 1.15 Rough endoplasmic reticulum is a system of membranes extending throughout the cytoplasm of a cell. It is covered with ribosomes the (tiny dots). Ribosomes are the site of protein synthesis.

There are thousands of different chemical reactions that take place inside cells. A key function of a cell membrane is to separate cell functions into different compartments so they don't take place together. For example, the reactions and enzymes of aerobic respiration are kept inside the mitochondria, separate from the rest of the cytoplasm.

## CHAPTER QUESTIONS

## SKILLS CRITICAL THINKING



More questions on life processes can be found at the end of Unit 1 on page 29.

1 Which of the following comparisons of animal and plant cells is *not* true?

	Animal cells	Plant cells
A	do not have chloroplasts	have chloroplasts
B	have mitochondria	do not have mitochondria
C	have temporary vacuoles	have permanent vacuoles
D	do not have cellulose cell walls	have cellulose cell walls

2 Which of the following descriptions is correct?

- A The cell wall is freely permeable and the cell membrane is partially permeable
- B The cell wall is partially permeable and the cell membrane is freely permeable
- C Both the cell wall and the cell membrane are freely permeable
- D Both the cell wall and the cell membrane are partially permeable



3 What are the products of anaerobic respiration in yeast?

- A ethanol and carbon dioxide
- B lactate and carbon dioxide
- C carbon dioxide and water
- D ethanol and water

## SKILLS INTERPRETATION

## SKILLS CRITICAL THINKING



4 a Draw a diagram of a plant cell. Label all of the parts. Alongside each label write the function of that part.

b Write down three differences between the cell you have drawn and a 'typical' animal cell.

5 Write a short description of the nature and function of enzymes. Include in your description:

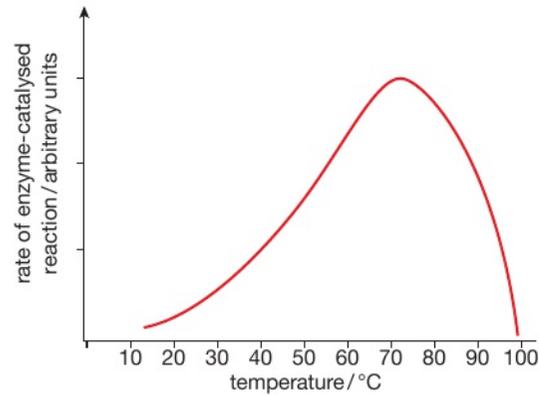
- a definition of an enzyme
- a description of the 'lock and key' model of enzyme action
- an explanation of the difference between intracellular and extracellular enzymes.

Your description should be about a page in length, including a labelled diagram.



6 The graph shows the effect of temperature on an enzyme. The enzyme was extracted from microorganism that lives in hot mineral springs near a volcano.

## SKILLS ANALYSIS



- a What is the optimum temperature of this enzyme?
- b Explain why the activity of the enzyme is greater at 60 °C than at 30 °C.
- c The optimum temperature of enzymes in the human body is about 37 °C. Explain why this enzyme is different.
- d What happens to the enzyme at 90 °C?

8<sup>th</sup>

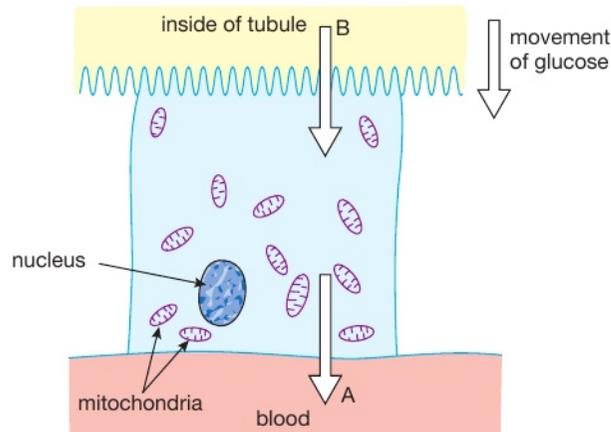
9<sup>th</sup>

8<sup>th</sup>

7<sup>th</sup>

6<sup>th</sup>

- 7 Explain the differences between diffusion and active transport.
- 8 The diagram shows a cell from the lining of a human kidney tubule. A major role of the cell is to absorb glucose from the fluid passing along the tubule and pass it into the blood, as shown by the arrow on the diagram.



- a What is the function of the mitochondria?
- b The tubule cell contains a large number of mitochondria. They are needed for the cell to transport glucose across the cell membrane into the blood at 'A'. Suggest the method that the cell uses to do this and explain your answer.
- c The mitochondria are *not* needed to transport the glucose into the cell from the tubule at 'B'. Name the process by which the ions move across the membrane at 'B' and explain your answer.
- d The surface membrane of the tubule cell at 'B' is greatly folded. Suggest how this adaptation helps the cell to carry out its function.

SKILLS INTERPRETATION

SKILLS CRITICAL THINKING, INTERPRETATION

SKILLS ANALYSIS, INTERPRETATION

5<sup>th</sup>

7<sup>th</sup>

6<sup>th</sup>

7<sup>th</sup>

## 2 THE VARIETY OF LIVING ORGANISMS

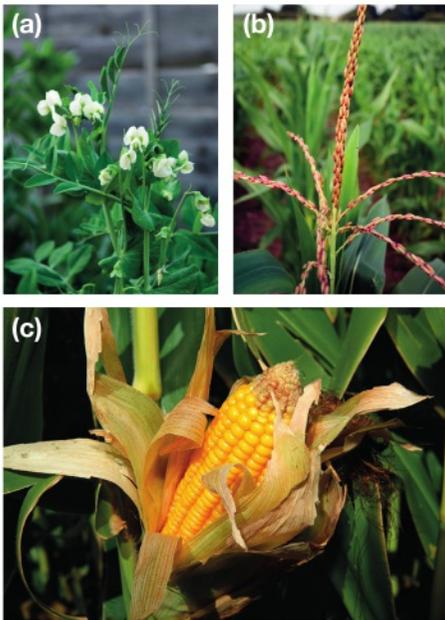
There is an enormous variety of living organisms. Biologists put them into groups according to their structure and function. The members of each group have certain features in common.

### LEARNING OBJECTIVES

- Understand the difference between eukaryotic and prokaryotic organisms
- Describe the features common to plants and recognise examples of flowering plants such as maize, peas and beans
- Describe the features common to animals and recognise examples such as mammals and insects
- Describe the features common to fungi and recognise examples such as *Mucor* and yeast
- Describe the features common to protocists and recognise examples such as *Amoeba*, *Chlorella* and *Plasmodium*
- Describe the features common to bacteria and recognise examples such as *Lactobacillus bulgaricus* and *Pneumococcus*
- Describe the features common to viruses and recognise examples such as the influenza virus, the HIV virus and the tobacco mosaic virus
- Understand the term 'pathogen' and know that pathogens may include fungi, bacteria, protocists or viruses.

There are more than ten million species of organisms alive on Earth today, and many more that once lived on Earth but are now extinct. In order to make sense of this enormous variety biologists classify organisms, putting them into groups. Members of each group are related – they are descended from a common ancestor by the process of evolution (see Chapter 19). This common ancestry is reflected in the similarities of structure and function of the members of a group.

The five major groups of living organisms are plants, animals, fungi, protocists and bacteria.



▲ Figure 2.1 (a) A pea plant. Its leaves and **stem cells** contain chloroplasts, giving them their green colour. The white flowers are pollinated by insects. (b) Maize plants are pollinated by wind. These are the male flowers, which make the pollen. (c) The female maize flowers produce seeds after pollination.

### PLANTS

You will be familiar with flowering plants, such as those shown in Figure 2.1. This group, or **kingdom**, also contains simpler plants, such as mosses and ferns. All plants are multicellular, which means that their 'bodies' are made up of many cells. Their main distinguishing feature is that their cells contain chloroplasts and they carry out photosynthesis – the process that uses light energy to convert simple inorganic molecules such as water and carbon dioxide into complex organic compounds (see Chapter 10). One of these organic compounds is the carbohydrate cellulose, and all plants have cell walls made of this material.

Plants can make many other organic compounds as a result of photosynthesis. One of the first to be made is the storage carbohydrate **starch**, which is often found inside plant cells. Another is the sugar **sucrose**, which is transported around the plant and is sometimes stored in fruits and other plant organs. The structure and function of flowering plants is dealt with in Unit 3 of this book.

## ANIMALS

You will be even more familiar with this kingdom, since it contains the species *Homo sapiens*, i.e. humans! The variety of the animal kingdom is also enormous, including organisms such as sponges, molluscs, worms, starfish, insects and crustaceans, through to larger animals such as fish, amphibians, reptiles, birds and mammals (Figure 2.2). The last five groups are all **vertebrates**, which means that they have a vertebral column, or backbone. All other animals lack this feature, and are called **invertebrates**.



▲ Figure 2.2 (a) A housefly. (b) A mosquito, feeding on human blood. Houseflies and mosquitoes are both insects, which make up the largest sub-group of all the animals. About 60% of all animal species are insects. (c) This high jumper's movement is coordinated by a complex nervous system.

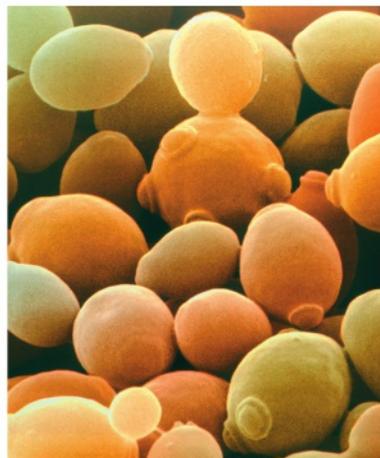
Animals are also multicellular organisms. Their cells never contain chloroplasts, so they are unable to carry out photosynthesis. Instead, they gain their nutrition by feeding on other animals or plants. Animal cells also lack cell walls, which allows their cells to change shape, an important feature for organisms that need to move from place to place. Movement in animals is achieved in various ways, but often involves coordination by a nervous system (see Chapter 6). Another feature common to most animals is that they store carbohydrate in their cells as a compound called **glycogen** (see Chapter 4). The structure and function of animals is dealt with in Unit 2 of this book.

## FUNGI

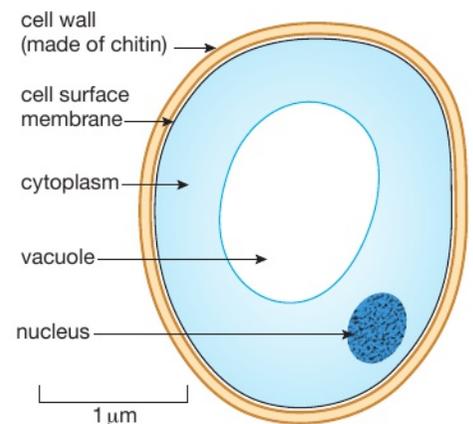
### EXTENSION WORK

Because fungi have cell walls, they were once thought to be plants that had lost their chlorophyll. We now know that their cell wall is not made of cellulose as in plants, but of a different chemical called **chitin** (the same material that makes up the outside skeleton of insects). There are many ways that fungi are very different from plants (the most obvious is that fungi do not photosynthesise) and they are not closely related to plants at all.

Fungi include mushrooms and toadstools, as well as moulds. These groups of fungi are multicellular. Another group of fungi is the yeasts, which are **unicellular** (made of single cells). Different species of yeasts live everywhere – on the surface of fruits, in soil, water, and even on dust in the air. The yeast powder used for baking contains millions of yeast cells (Figure 2.3). The cells of fungi never contain chloroplasts, so they cannot photosynthesise. Their cells have cell walls, but they are not composed of cellulose (Figure 2.4).



▲ Figure 2.3 Yeast cells, highly magnified



▲ Figure 2.4 Structure of a yeast cell

## KEY POINT

The singular of hyphae is hypha.



▲ Figure 2.5 Toadstools growing on a rotting tree trunk

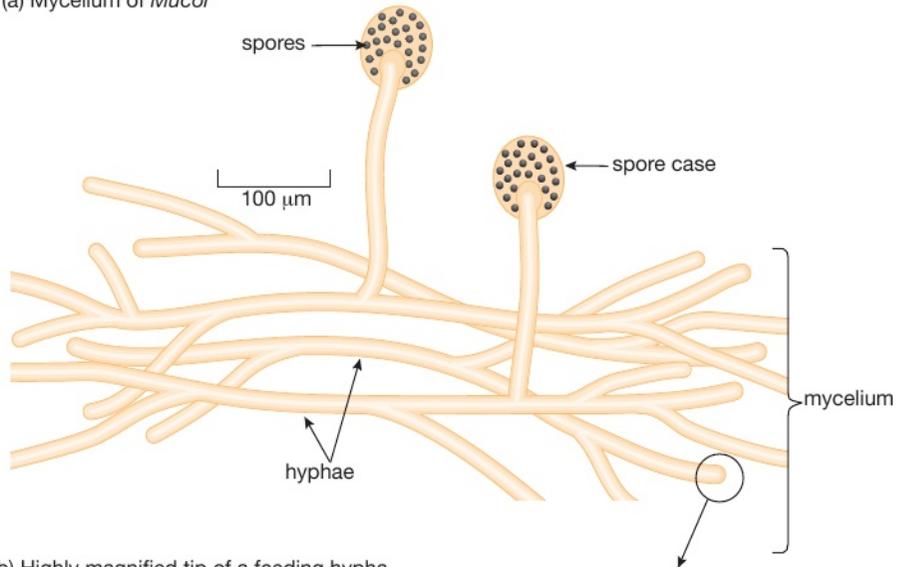


▲ Figure 2.6 The 'pin mould' *Mucor* growing on a piece of bread. The dark spots are structures that produce spores for reproduction.

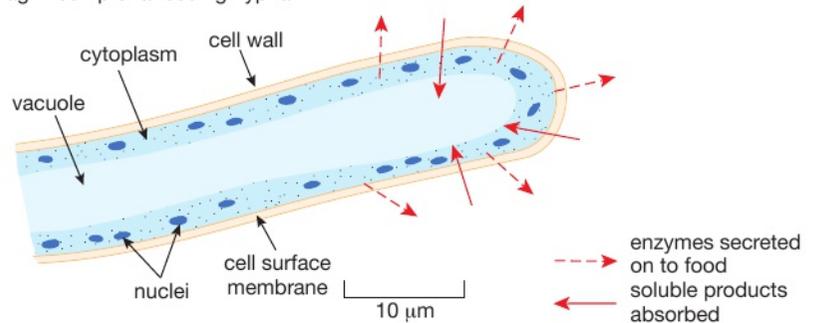
A mushroom or toadstool is the reproductive structure of the organism, called a fruiting body (Figure 2.5). Under the soil, the mushroom has many fine thread-like filaments called **hyphae** (pronounced high-fee). A mould is rather like a mushroom without the fruiting body. It just consists of the network of hyphae (Figure 2.6). The whole network is called a **mycelium** (pronounced my-sea-lee-um). Moulds feed by absorbing nutrients from dead (or sometimes living) material, so they are found wherever this is present, for example, in soil, rotting leaves or decaying fruit.

If you leave a piece of bread or fruit exposed to the air for a few days, it will soon become mouldy. Mould spores carried in the air have landed on the food and grown into a mycelium of hyphae (Figure 2.7).

(a) Mycelium of *Mucor*



(b) Highly magnified tip of a feeding hypha



▲ Figure 2.7 The structure of a typical mould fungus, the 'pin mould' *Mucor*.

The thread-like hyphae of *Mucor* have cell walls surrounding their cytoplasm. The cytoplasm contains many nuclei. In other words the hyphae are not divided up into separate cells.

When a spore from *Mucor* lands on the food, a hypha grows out from it. The hypha grows and branches again and again, until the mycelium covers the surface of the food. The hyphae secrete digestive enzymes on to the food, breaking it down into soluble substances such as sugars, which are then absorbed by the mould. Eventually, the food is used up and the mould must infect another source of food by producing more spores.

When an organism feeds on dead organic material in this way, and digestion takes place outside of the organism, this is called **saprotrophic** nutrition. Enzymes that are secreted out of cells for this purpose are called **extracellular** enzymes (see Chapter 1).

## PROTOCTISTS

**Protoctists** are sometimes called the ‘dustbin kingdom’, because they are a mixed group of organisms that don’t fit into the plants, animals or fungi. Most protoctists are microscopic single-celled organisms (Figure 2.8). Some look like animal cells, such as *Amoeba*, which lives in pond water. These are known as **protozoa**. Other protoctists have chloroplasts and carry out photosynthesis, so are more like plants. These are called **algae**. Most algae are unicellular, but some species such as seaweeds are multicellular and can grow to a great size. Some protoctists are the agents of disease, such as *Plasmodium*, the organism that causes malaria.



▲ Figure 2.8 (a) *Amoeba*, a protozoan that lives in ponds (b) *Chlorella*, a unicellular freshwater alga (c) Blood cells containing the protoctist parasite *Plasmodium*, the organism responsible for causing malaria

## EUKARYOTIC AND PROKARYOTIC ORGANISMS

All the organisms described so far are composed of **eukaryotic** cells and are known as eukaryotic organisms. ‘Eukaryotic’ means ‘having a nucleus’ – their cells contain a nucleus surrounded by a membrane, along with other membrane bound organelles, such as mitochondria and chloroplasts.

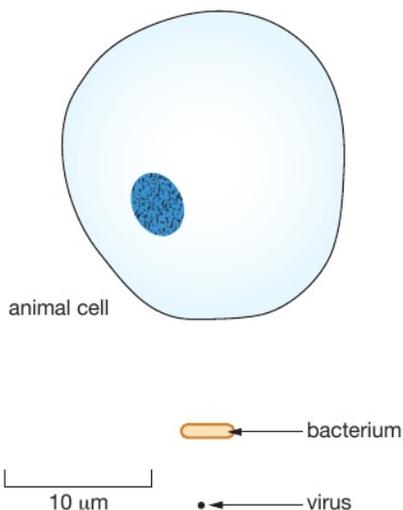
There are also organisms made of simpler cells, which have no nucleus, mitochondria or chloroplasts. These are called **prokaryotic** cells. ‘Prokaryotic’ means ‘before nucleus’. The main forms of prokaryotic organisms are the bacteria.

## BACTERIA

**Bacteria** are small single-celled organisms. Their cells are much smaller than those of eukaryotic organisms and have a much simpler structure. To give you some idea of their size, a typical animal cell might be 10 to 50  $\mu\text{m}$  in diameter (1  $\mu\text{m}$ , or one micrometre, is a millionth of a metre). Compared with this, a typical bacterium is only 1 to 5  $\mu\text{m}$  in length (Figure 2.9) and its volume is thousands of times smaller than that of the animal cell.

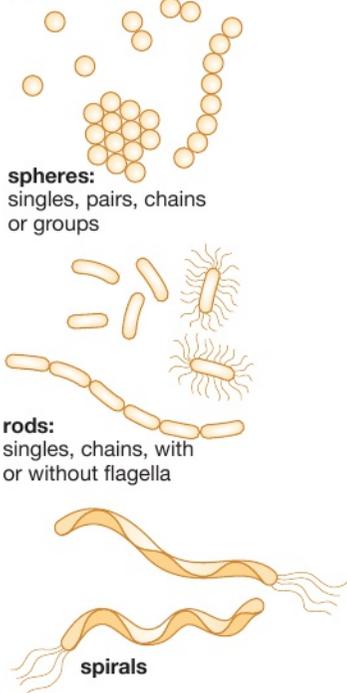
There are three basic shapes of bacteria: spheres, rods and spirals, but they all have a similar internal structure (Figure 2.10).

All bacteria are surrounded by a cell wall, which protects the bacterium and keeps the shape of the cell. Bacterial cell walls are not made of cellulose but a complex compound of sugars and proteins called peptidoglycan. Some species have another layer outside this wall, called a **capsule** or slime layer. Both give the bacterium extra protection. Underneath the cell wall is the cell membrane, as in other cells. The middle of the cell is made of cytoplasm. Since it is a prokaryotic cell, the bacterium has no nucleus. Instead, its genetic material (DNA) is in a single chromosome, loose in the cytoplasm, forming a circular loop.

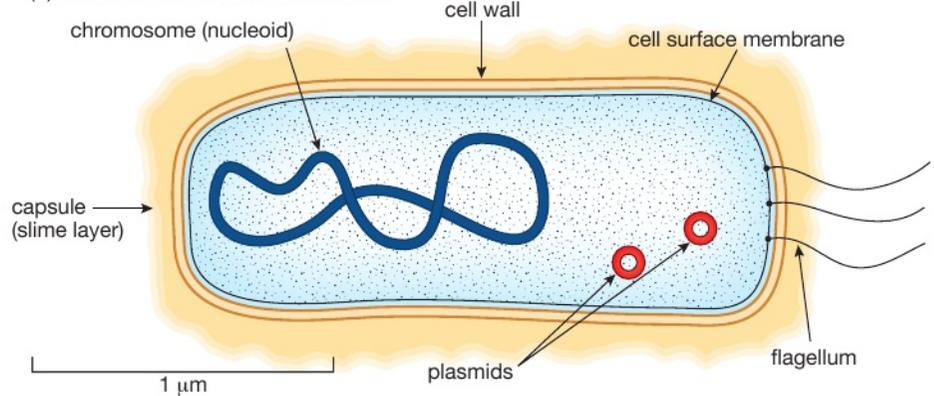


▲ Figure 2.9 A bacterium is much smaller than an animal cell. The relative size of a virus is also shown.

(a) Some different bacterial shapes



(b) Internal structure of a bacterium



▲▲ Figure 2.10 Structure of bacteria

Some bacteria can swim, and are propelled through water by corkscrew-like movements of structures called flagella (a single one of these is called a **flagellum**). However, many bacteria do not have flagella and cannot move by themselves. Other structures present in the cytoplasm include the **plasmids**. These are small circular rings of DNA, carrying some of the bacterium's genes. Not all bacteria contain plasmids, although about three-quarters of all known species do. Plasmids have very important uses in genetic engineering (see Chapter 21).

Some bacteria contain a form of chlorophyll in their cytoplasm, and can carry out photosynthesis. However, most bacteria feed off other living or dead organisms. Along with the fungi, many bacteria are important **decomposers** (see Chapter 14), recycling dead organisms and waste products in the soil and elsewhere. Some bacteria are used by humans to make food, such as *Lactobacillus bulgaricus*, a rod-shaped species used in the production of yoghurt from milk (Figure 2.11). Other species are **pathogens**, which means that they cause disease (Figure 2.12).

Despite the relatively simple structure of the bacterial cell, it is still a living cell that carries out the normal processes of life, such as respiration, feeding, excretion, growth and reproduction. As you have seen, some bacteria can move, and they can also respond to a range of stimuli. For example, they may move towards a source of food, or away from a poisonous chemical. You should think about these features when you compare bacteria with the next group, the much simpler viruses.

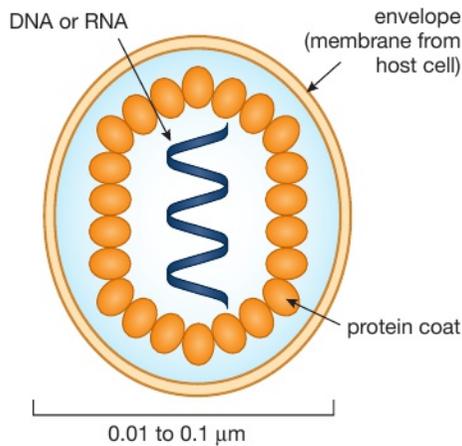
**KEY POINT**

'Pathogens are organisms that cause disease. Many common animal and plant diseases are caused by bacteria or viruses. Most protists are free-living, but a few species are pathogens, such as *Plasmodium* (Figure 2.8). Even some species of fungi can cause disease, e.g. the skin infection called 'athlete's foot' is caused by a mould.

▲ Figure 2.11 The bacterium *Lactobacillus bulgaricus*, used in the production of yoghurt.▲ Figure 2.12 Rounded cells of the bacterium *Pneumococcus*, one cause of pneumonia.**VIRUSES**

All **viruses** are **parasites**, and can only reproduce inside living cells. The cell in which the virus lives is called the host. There are many different types of viruses. Some live in the cells of animals or plants, and there are even viruses which infect bacteria. Viruses are much smaller than bacterial cells: most are between 0.01 and 0.1  $\mu\text{m}$  in diameter (Figure 2.9).

Viruses are not made of cells. A virus particle is very simple. It has no nucleus or cytoplasm, and is composed of a core of genetic material surrounded by a protein coat (Figure 2.13). The genetic material can be either **DNA**, or a similar chemical called **ribonucleic acid (RNA)**. In either case, the genetic material makes up just a few genes – all that is needed for the virus to reproduce inside its host cell.



▲ Figure 2.13 The structure of a typical virus, such as the type causing influenza (flu).

### EXTENSION WORK

AIDS is not actually a disease but a 'syndrome'. A syndrome is a set of symptoms caused by a medical condition. In the case of HIV, the virus severely damages the person's immune system, so they are more likely to get other diseases, such as tuberculosis. They may also develop some unusual types of cancer. This collection of different symptoms is referred to as AIDS.



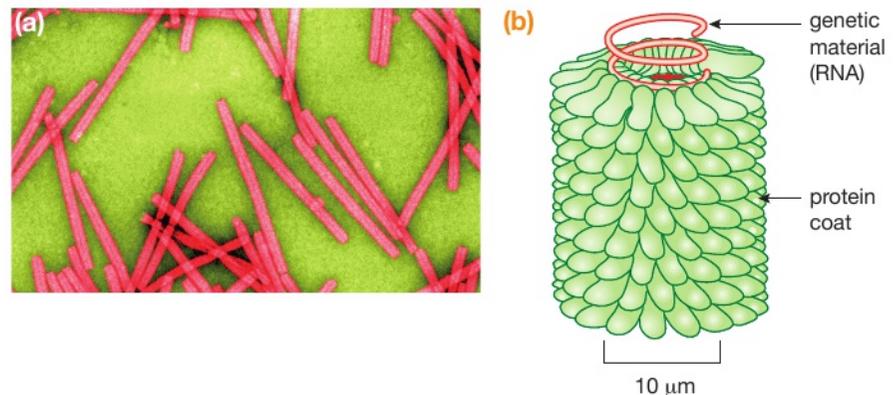
▲ Figure 2.15 Discolouration of the leaves of a tobacco plant, caused by infection with tobacco mosaic virus.

Sometimes a membrane called an envelope may surround a virus particle, but the virus does not make this. Instead it is 'stolen' from the surface membrane of the host cell.

Viruses do not feed, respire, excrete, move, grow or respond to their surroundings. They do not carry out any of the normal 'characteristics' of living things except reproduction, and they can only do this parasitically. This is why biologists do not consider viruses to be living organisms. You can think of them as being on the border between an organism and a non-living chemical.

A virus reproduces by entering the host cell and taking over the host's genetic machinery to make more virus particles. After many virus particles have been made, the host cell dies and the particles are released to infect more cells. Many human diseases are caused in this way, such as influenza ('flu'). Other examples include colds, measles, mumps, polio and rubella ('German measles'). Of course, the reproduction process does not continue forever. Usually, the body's immune system destroys the virus and the person recovers. Sometimes, however, a virus cannot be destroyed by the immune system quickly enough, and it may cause permanent damage or death. With other infections, the virus may attack cells of the immune system itself. This is the case with HIV (the Human Immunodeficiency Virus), which causes the illness called AIDS (Acquired Immune Deficiency Syndrome).

Viruses don't just parasitise animal cells. Some infect plant cells, such as the tobacco mosaic virus (Figure 2.14), which interferes with the ability of the tobacco plant to make chloroplasts, causing mottled patches to develop on the leaves (Figure 2.15).



▲ Figure 2.14 (a) Tobacco mosaic virus (TMV), seen through an electron microscope. (b) Structure of part of a TMV particle, magnified 1.25 million times.

## CHAPTER QUESTIONS

More questions on the variety of living organisms can be found at the end of Unit 1 on page 29.

### SKILLS CRITICAL THINKING



1 Which of the following is *not* a characteristic of plants?

- A cells contain chloroplasts
- B cell wall made of cellulose
- C bodies are multicellular
- D store carbohydrate as glycogen

## SKILLS CRITICAL THINKING



- 2 Fungi carry out *saprotrophic nutrition*. What is the meaning of this term?
- A** extracellular digestion of dead organic matter  
**B** feeding on other living organisms  
**C** making organic molecules by photosynthesis  
**D** secreting digestive enzymes



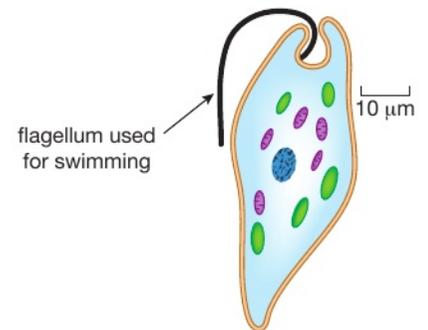
- 3 Below are three groups of organisms.
1. viruses
  2. bacteria
  3. yeasts

Which of these organisms are prokaryotic?

- A** 1 only  
**B** 2 only  
**C** 1 and 2  
**D** 1, 2 and 3
- 4 Which of the following diseases is *not* caused by a virus?
- A** influenza  
**B** measles  
**C** malaria  
**D** AIDS
- 5 a Name the kingdom to which each of the following organisms belongs:
- i mushroom
  - ii *Chlorella*
  - iii moss
  - iv *Lactobacillus*

## SKILLS ANALYSIS, REASONING

- b The diagram shows a species of protist called *Euglena*. Use the diagram to explain why *Euglena* is classified as a protist and not as an animal or plant.



## SKILLS INTERPRETATION



- 6 a Draw a diagram to show the structure of a typical virus particle.

## SKILLS CRITICAL THINKING



- b Is a virus a living organism? Explain your answer.  
**c** Explain the statement 'viruses are all parasites'.



- 7 Explain the meanings of the following terms:



- a** invertebrate  
**b** hyphae  
**c** saprotrophic

# UNIT QUESTIONS

**SKILLS** CRITICAL THINKING

**1**

These three organelles are found in cells: nucleus, chloroplast and mitochondrion.

- a** Which of the above organelles would be found in:
- i** a cell from a human muscle? (1)
  - ii** a palisade cell from a leaf? (1)
  - iii** a cell from the root of a plant? (1)
- b** Explain fully why the answers to ii) and iii) above are different. (1)
- c** What is the function of each organelle? (3)

**(Total 7 marks)**

**SKILLS** REASONING

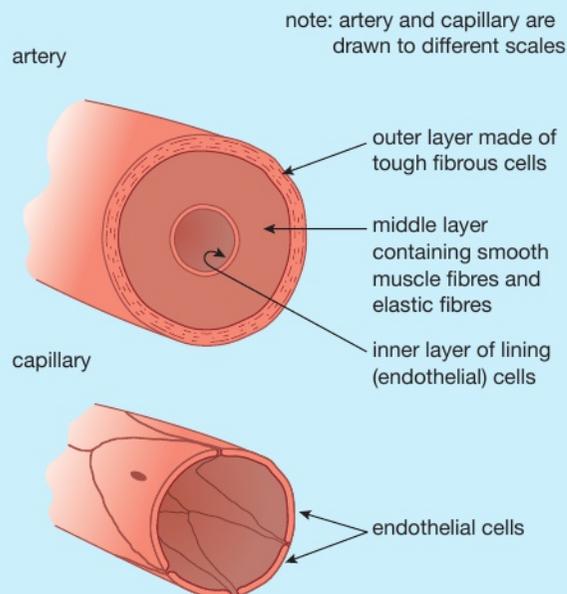
**SKILLS** INTERPRETATION

**SKILLS** INTERPRETATION, REASONING

**2**

In multicellular organisms, cells are organised into tissues, organs and organ systems.

- a** The diagram shows a section through an artery and a capillary.



Explain why an artery can be considered to be an organ whereas a capillary cannot. (2)

- b** Organ systems contain two or more organs whose functions are linked. The digestive system is one human organ system. (See Chapter 4.)
- i** What does the digestive system do? (2)
  - ii** Name three organs in the human digestive system. Explain what each organ does as part of the digestive system. (6)
  - iii** Name two other human organ systems and, for each system, name two organs that are part of the system. (6)

**(Total 16 marks)**

**SKILLS** CRITICAL THINKING


## SKILLS ANALYSIS

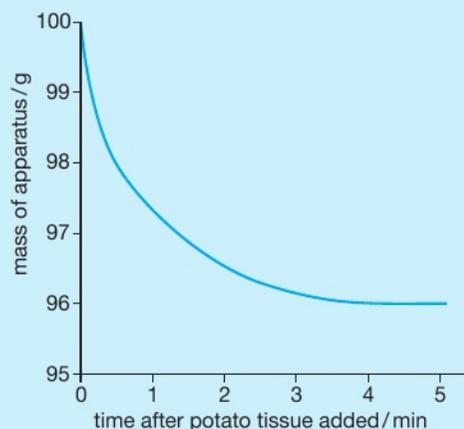


3

Catalase is an enzyme found in many plant and animal cells. It catalyses the breakdown of hydrogen peroxide into water and oxygen.



- a In an investigation into the action of catalase in potato, 20 g of potato tissue was put into a small beaker containing hydrogen peroxide weighing 80 g in total. The temperature was maintained at 20 °C throughout the investigation. As soon as the potato was added, the mass of the beaker and its contents was recorded until there was no further change in mass. The results are shown in the graph.



## SKILLS REASONING

- i How much oxygen was formed in this investigation? Explain your answer. (2)
- ii Estimate the time by which half this mass of oxygen had been formed. (2)
- iii Explain, in terms of collisions between enzyme and substrate molecules, why the rate of reaction changes during the course of the investigation. (2)

- b The students repeated the investigation at 30 °C. What difference, if any, would you expect in:

- i the mass of oxygen formed?
- ii the time taken to form this mass of oxygen?
- Explain your answers. (4)

**(Total 10 marks)**

## CRITICAL THINKING



4

Different particles move across cell membranes using different processes.

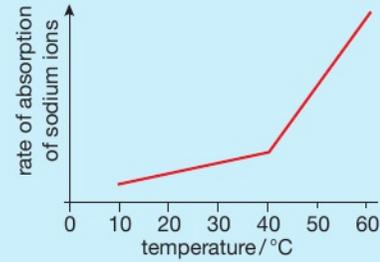
- a The table below shows some ways in which active transport, osmosis and diffusion are similar and some ways in which they are different. Copy and complete the table with ticks and crosses. (3)

Feature	Active transport	Osmosis	Diffusion
particles must have kinetic energy			
requires energy from respiration			
particles move down a concentration gradient			

SKILLS CRITICAL THINKING



**b** The graph shows the results of an investigation into the rate of diffusion of sodium ions across the membranes of potato cells.



- i** Explain the increase in the rate of diffusion up to 40 °C. (2)
- ii** Suggest why the rate of increase is much steeper at temperatures above 40 °C. (2)

**(Total 7 marks)**

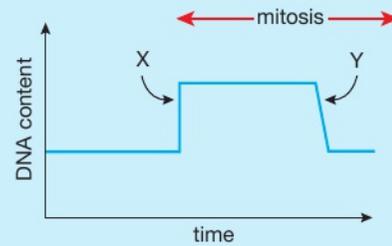
SKILLS REASONING



**5**

Cells in the wall of the small intestine divide by mitosis to replace cells lost as food passes through.

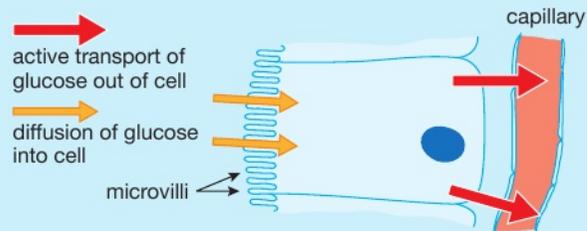
**a** Chromosomes contain DNA. The graph shows the changes in the DNA content of a cell in the wall of the small intestine as it divides by mitosis.



- i** Why is it essential that the DNA content is doubled (X) before mitosis begins? (2)
- ii** What do you think happens to the cell at point Y? (1)

SKILLS ANALYSIS

**b** The diagram shows a cell in the wall of a villus in the small intestine. Some of the processes involved in the absorption of glucose are also shown.



SKILLS INTERPRETATION



- i** What is the importance of the small intestine having villi? (1)
- ii** Suggest how the microvilli adapt this cell to its function of absorbing glucose. (1)
- iii** Suggest how the active transport of glucose out of the cell and into the blood stream helps with the absorption of glucose from the small intestine. (2)

**(Total 7 marks)**

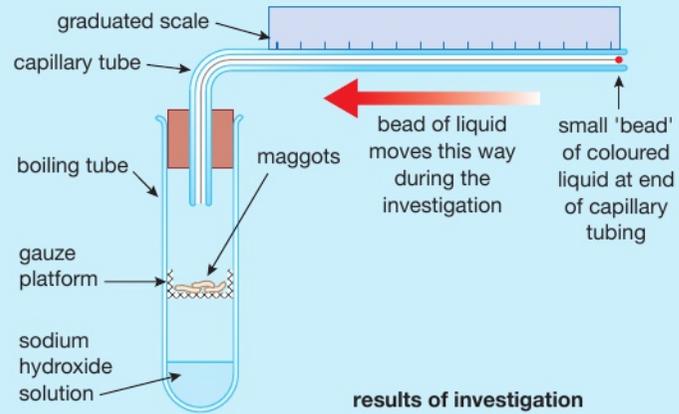
## SKILLS REASONING



6

A respirometer is used to measure the rate of respiration. The diagram shows a simple respirometer. The sodium hydroxide solution in the apparatus absorbs carbon dioxide. Some results from the investigation are also shown.

**Safety note:** Eye protection should be worn when setting up the apparatus as sodium hydroxide is very hazardous to the eyes.



Experiment	Distance moved by bead / mm
1	20
2	3
3	18

Assume that the maggots in the apparatus respire aerobically.

- Write the symbol equation for aerobic respiration. (4)
- From the equation, what can you assume about the amount of oxygen taken in and carbon dioxide given off by the maggots? Explain your answer. (3)
- Result 2 is significantly different from the other two results. Suggest a reason for this. (2)
- How would the results be different if the organisms under investigation respired anaerobically? (2)

**(Total 11 marks)**



## SKILLS INTERPRETATION



7

The table below shows some features of different groups of organisms. Copy and complete the table by putting a tick in the box if the organism has that feature, or a cross if it lacks the feature.

Feature	Type of organism		
	Plant	Fungus	Virus
they are all parasites			
they are made up of a mycelium of hyphae			
they can only reproduce inside living cells			
they feed by extracellular digestion by enzymes			
they store carbohydrate as starch			

**(Total 5 marks)**

## SKILLS CRITICAL THINKING



8

Copy and complete the following account.

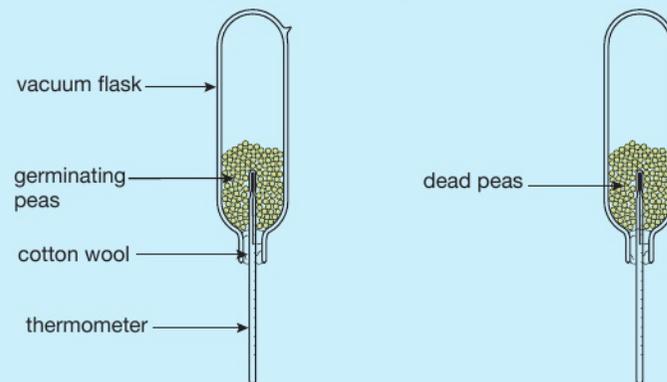
Plants have cell walls made of \_\_\_\_\_. They store carbohydrate as the insoluble compound called \_\_\_\_\_ or sometimes as the sugar \_\_\_\_\_. Plants make these substances as a result of the process called \_\_\_\_\_. Animals, on the other hand, store carbohydrate as the compound \_\_\_\_\_. Both animals' and plants' cells have nuclei, but the cells of bacteria lack a true nucleus, having their DNA in a circular chromosome. They sometimes also contain small rings of DNA called \_\_\_\_\_, which are used in genetic engineering. Bacteria and fungi break down organic matter in the soil. They are known as \_\_\_\_\_. Some bacteria are pathogens, which means that they \_\_\_\_\_.

**(Total 8 marks)**

## SKILLS DECISION MAKING

9

The diagram shows the apparatus used to investigate germination of pea seeds. At the start of the experiment the temperature in both flasks was 19 °C



The apparatus was left for 24 hours. At the end of this time the temperature in the flask with germinating peas was 22 °C, while the temperature in the flask with dead peas was 19 °C.

- Explain the biological reason for this difference in temperature. (2)
- The seeds in both flasks were washed in disinfectant before the experiment. Explain why this was done. (1)
- Cotton wool was used to hold the seeds in the flasks. Suggest why cotton wool was used instead of a rubber bung. (1)
- Pea seeds were used in both flasks. State another variable that should have been controlled in the experiment. (1)

**(Total 5 marks)**

## SKILLS EXECUTIVE FUNCTION



10

A piece of meat is a tissue composed of muscle fibres. Muscle fibres use ATP when they contract. Describe an investigation to find out if a solution of ATP will cause the contraction of muscle fibres.

**(Total 6 marks)**

BREATHING AND GAS EXCHANGE 35

FOOD AND DIGESTION 48

BLOOD AND CIRCULATION 64

COORDINATION 77

CHEMICAL COORDINATION 91

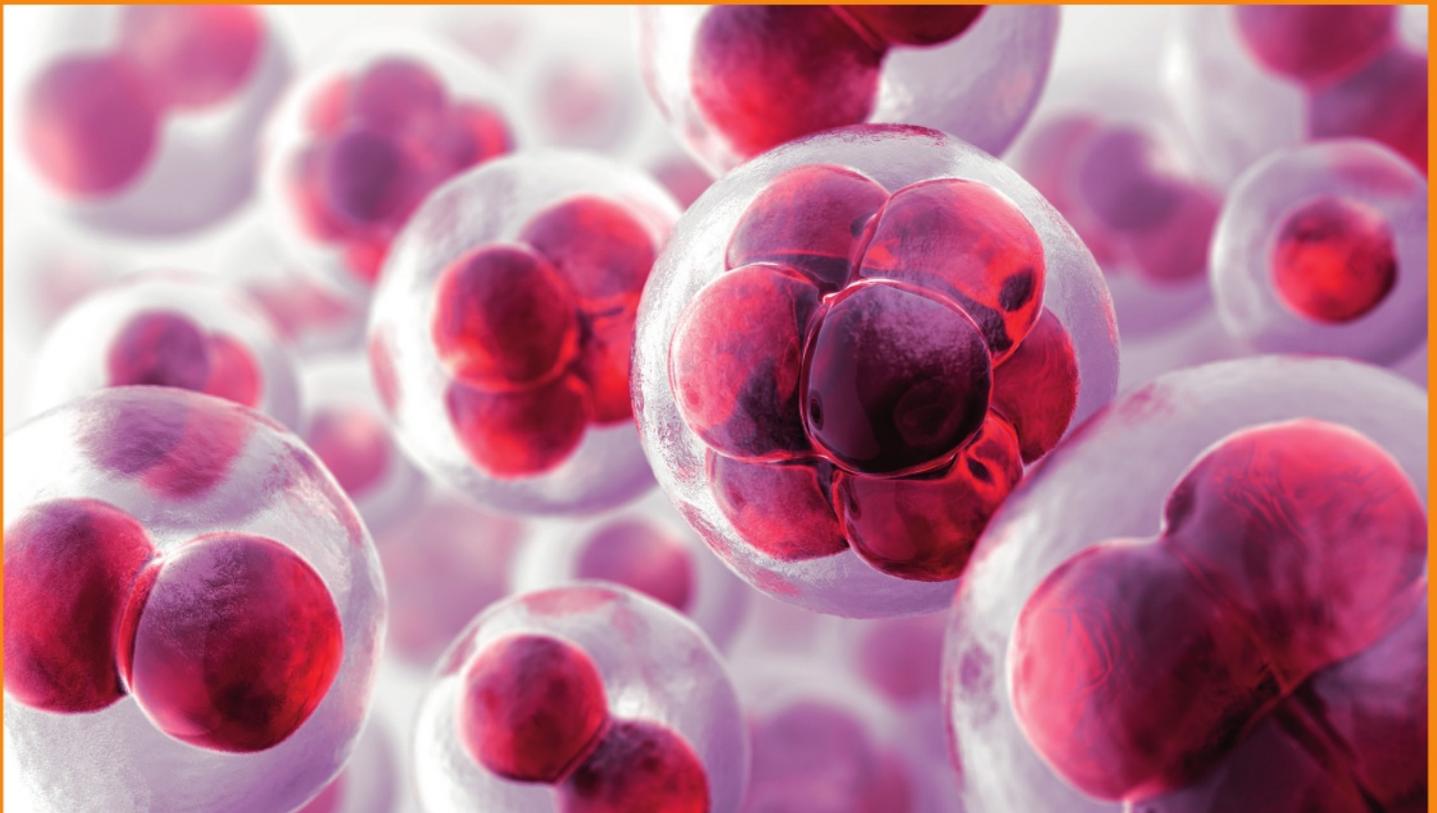
HOMEOSTASIS AND EXCRETION 97

REPRODUCTION IN HUMANS 104

# BIOLOGY UNIT 2

# ANIMAL PHYSIOLOGY

Physiology is the branch of biology that looks at how living things function. It studies the workings of an organism at different levels from cells, tissues and organs through to the whole organism. In this unit we look at animal physiology and in particular how the human body works. It is important to study physiology, not least because knowledge of the body is essential in understanding how to treat it when it goes wrong. This is reflected in the fact that one of the six categories of Nobel Prize is awarded for 'physiology or medicine'.



## 3 BREATHING AND GAS EXCHANGE

When we breathe, air is moved in and out of the lungs so that gas exchange can take place between the air and the blood. This chapter looks at these processes, and also deals with some ways that smoking can damage the lungs and stop these vital organs from working properly.

### LEARNING OBJECTIVES

- Describe the structure of the thorax, including the ribs, intercostal muscles, diaphragm, trachea, bronchi, bronchioles, alveoli and pleural membranes
- Understand the role of the intercostal muscles and the diaphragm in ventilation
- Explain how alveoli are adapted for gas exchange by diffusion between air in the lungs and blood in capillaries
- Investigate breathing in humans, including the release of carbon dioxide and the effect of exercise
- Understand the biological consequences of smoking in relation to the lungs and circulatory system, including coronary heart disease.

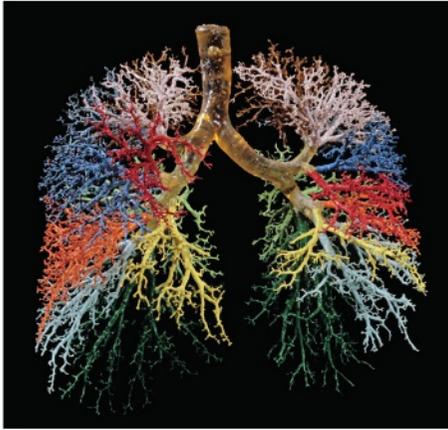
Cells get their energy by oxidising foods such as glucose, during the process called **respiration**. If cells are to respire aerobically, they need a continuous supply of oxygen from the blood. In addition, carbon dioxide from respiration needs to be removed from the body. In humans, these gases are exchanged between the blood and the air in the lungs.

### RESPIRATION AND BREATHING

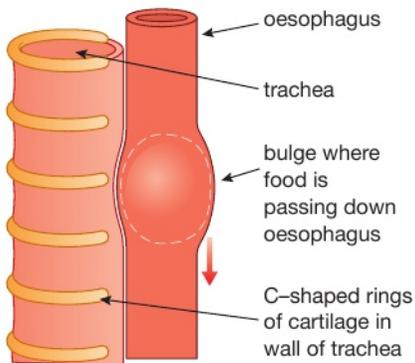
You need to understand the difference between respiration and breathing. Respiration is the oxidation reaction that releases energy from foods such as glucose (Chapter 1). Breathing is the mechanism that moves air into and out of the lungs, allowing gas exchange to take place. The lungs and associated structures are often called the 'respiratory system' but this can be confusing. It is better to call them the gas exchange system and this is the term we use in this book.

### THE STRUCTURE OF THE GAS EXCHANGE SYSTEM

The lungs are enclosed in the chest or **thorax** by the ribcage and a muscular sheet of tissue called the **diaphragm** (Figure 3.1). As you will see, the actions of these two structures bring about the movements of air into and out of the lungs. Joining each rib to the next are two sets of muscles called **intercostal muscles** ('costals' are rib bones). The diaphragm separates the contents of the thorax from the abdomen. It is not flat, but a shallow dome shape, with a fibrous middle part forming the 'roof' of the dome, and muscular edges forming the walls.



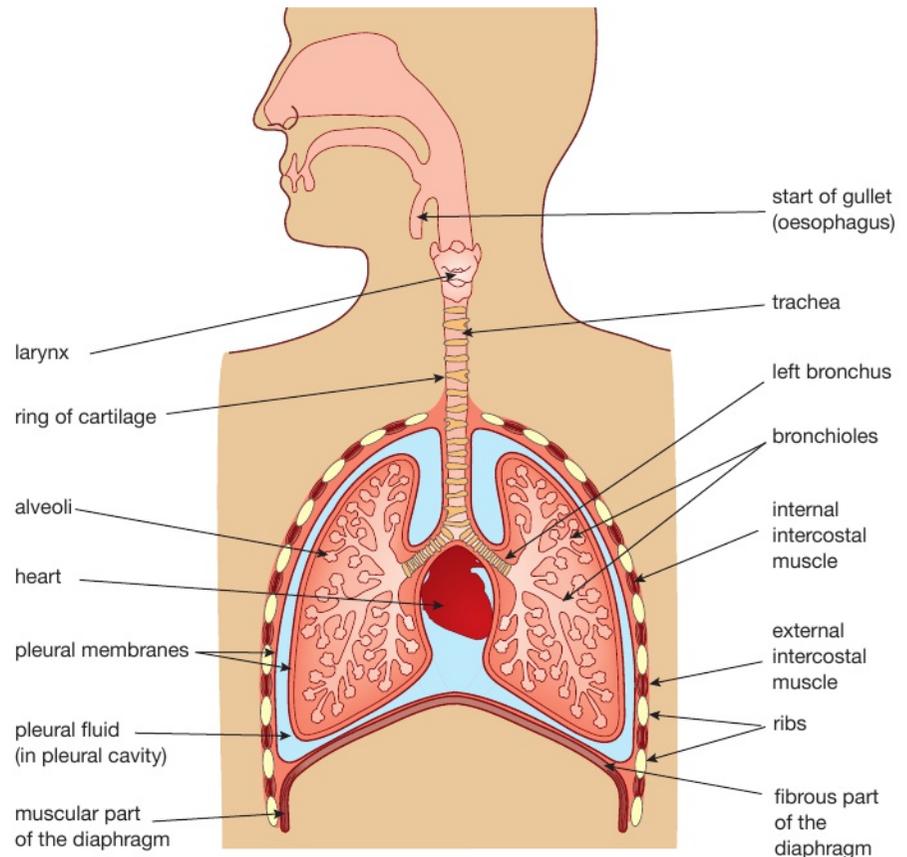
▲ Figure 3.2 This cast of the human lungs was made by injecting a pair of lungs with a liquid plastic. The plastic was allowed to set, then the lung tissue was dissolved away with acid.



▲ Figure 3.3 C-shaped cartilage rings in the trachea.

#### EXTENSION WORK

In the bronchi, the cartilage forms complete, circular rings. In the trachea, the rings are incomplete, and shaped like a letter 'C'. The open part of the ring is at the back of the trachea, next to where the oesophagus (gullet) lies as it passes through the thorax. When food passes along the oesophagus by peristalsis (see Chapter 4) the gaps in the rings allow the lumps of food to pass through more easily, without the peristaltic wave 'catching' on the rings (Figure 3.3).



▲ Figure 3.1 The human gas exchange system

The air passages of the lungs form a highly branching network (Figure 3.2). This is why it is sometimes called the **bronchial tree**.

When we breathe in, air enters our nose or mouth and passes down the windpipe or **trachea**. The trachea splits into two tubes called the **bronchi** (singular **bronchus**), one leading to each lung. Each bronchus divides into smaller and smaller tubes called **bronchioles**, eventually ending at microscopic air sacs, called **alveoli** (singular **alveolus**). It is here that gas exchange with the blood takes place.

The walls of trachea and bronchi contain rings of gristle or **cartilage**. These support the airways and keep them open when we breathe in. They are rather like the rings in a vacuum cleaner hose – without them the hose would squash flat when the cleaner sucks air in.

The inside of the thorax is separated from the lungs by two thin, moist membranes called the **pleural membranes**. They make up a continuous envelope around the lungs, forming an airtight seal. Between the two membranes is a space called the **pleural cavity**, filled with a thin layer of liquid called **pleural fluid**. This acts as lubrication, so that the surfaces of the lungs don't stick to the inside of the chest wall when we breathe.

#### KEEPING THE AIRWAYS CLEAN

The trachea and larger airways are lined with a layer of cells that have an important role in keeping the airways clean. Some cells in this lining secrete a sticky liquid called **mucus**, which traps particles of dirt or bacteria that are breathed in. Other cells are covered with tiny hair-like structures called **cilia** (Figure 3.4). The cilia beat backward and forward, sweeping the mucus and trapped particles out towards the mouth. In this way, dirt and bacteria are



▲ Figure 3.4 This electron microscope picture shows cilia from the lining of the trachea.

#### EXTENSION WORK

During normal (shallow) breathing, the elasticity of the lungs and the weight of the ribs acting downwards is enough to cause exhalation. The internal intercostals are only really used for deep (forced) breathing out, for instance when we are exercising.

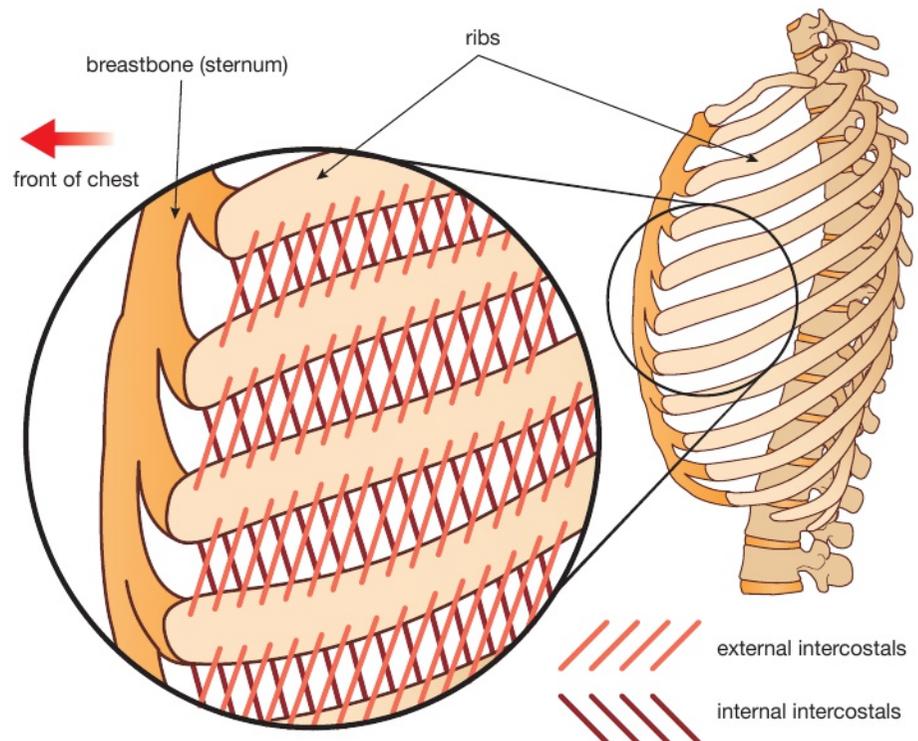
prevented from entering the lungs, where they might cause an infection. As you will see, one of the effects of smoking is that it destroys the cilia and stops this protection mechanism from working properly.

## VENTILATION OF THE LUNGS

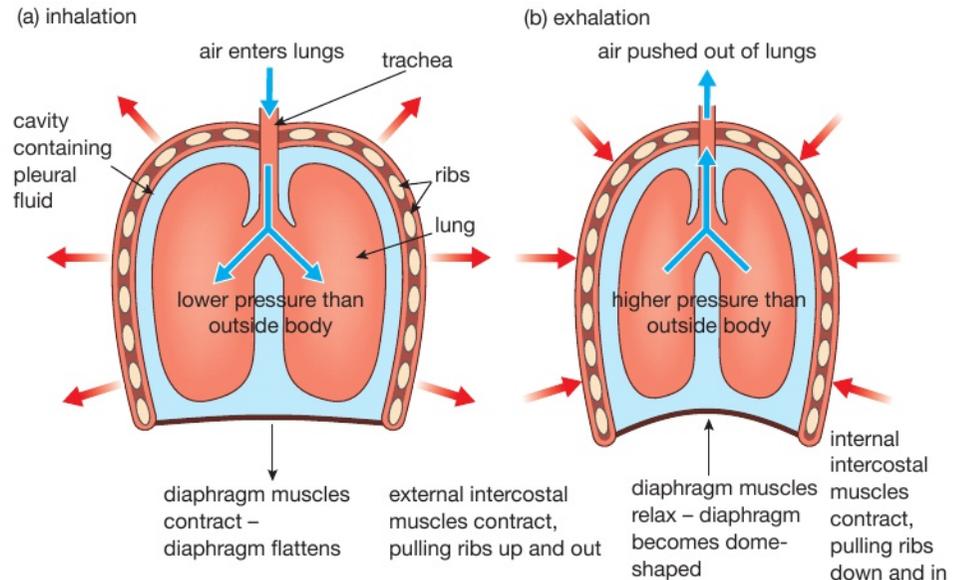
**Ventilation** means moving air in and out of the lungs. This requires a difference in air pressure – the air moves from a place where the pressure is high to one where it is low. Ventilation depends on the fact that the thorax is an airtight cavity. When we breathe, we change the volume of our thorax, which alters the pressure inside it. This causes air to move in or out of the lungs.

There are two movements that bring about ventilation: those of the ribs and the diaphragm. If you put your hands on your chest and breathe in deeply, you can feel your ribs move upwards and outwards. They are moved by the intercostal muscles (Figure 3.5). The outer (external) intercostals contract, pulling the ribs up. At the same time, the muscles of the diaphragm contract, pulling the diaphragm down into a more flattened shape (Figure 3.6a). Both these movements increase the volume of the chest and cause a slight drop in pressure inside the thorax compared with the air outside. Air then enters the lungs (inhalation).

The opposite happens when you breathe out deeply. The external intercostals relax, and the internal intercostals contract, pulling the ribs down and in. At the same time, the diaphragm muscles relax and the diaphragm goes back to its normal dome shape. The volume of the thorax decreases, and the pressure in the thorax is raised slightly above atmospheric pressure. This time the difference in pressure forces air out of the lungs (Figure 3.6b). Exhalation is helped by the fact that the lungs are elastic, so that they have a tendency to collapse and empty like a balloon.



▲ Figure 3.5 Side view of the chest wall, showing the ribs. The diagram shows how the two sets of intercostal muscles run between the ribs. When the external intercostals contract, they move the ribs upwards. When the internal intercostals contract, the ribs are moved downwards.



▲ Figure 3.6 Changes in the position of the ribs and diaphragm during breathing. (a) Breathing in (inhalation). (b) Breathing out (exhalation).

#### KEY POINT

It is important that you remember the changes in volume and pressure during ventilation. If you have trouble understanding these, think of what happens when you use a bicycle pump. If you push the pump handle, the air in the pump is squashed, its pressure rises and it is forced out of the pump. If you pull on the handle, the air pressure inside the pump falls a little, and air is drawn in from outside. This is similar to what happens in the lungs. In exams, students sometimes talk about the lungs *forcing* the air in and out – they don't!

## GAS EXCHANGE IN THE ALVEOLI

You can tell what is happening during gas exchange if you compare the amounts of different gases in atmospheric air with the air breathed out (Table 3.1).

Table 3.1 Approximate percentage volume of gases in atmospheric (inhaled) and exhaled air.

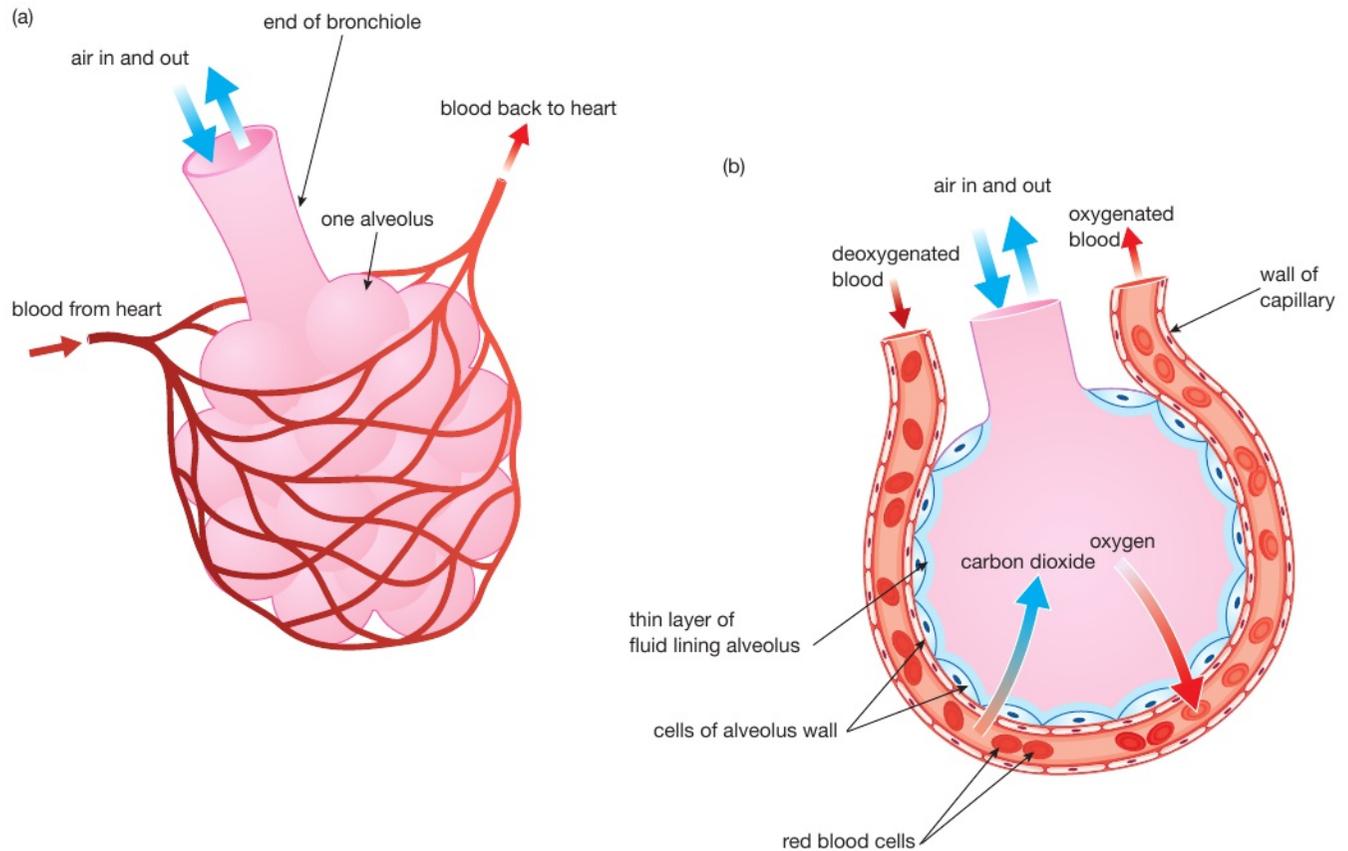
Gas	Atmospheric air / %	Exhaled air / %
nitrogen	78	79
oxygen	21	16
carbon dioxide	0.04	4
other gases (mainly argon)	1	1

Exhaled air is also warmer than atmospheric air, and is saturated with water vapour. The amount of water vapour in the atmosphere varies depending on weather conditions.

Clearly, the lungs are absorbing oxygen into the blood and removing carbon dioxide from it. This happens in the alveoli. To do this efficiently, the alveoli must have a structure which brings the air and blood very close together, over a very large surface area. There are enormous numbers of alveoli. It has been calculated that the two lungs contain about 700 000 000 of these tiny air sacs, giving a total surface area of 60 m<sup>2</sup>. That's bigger than the floor area of an average classroom! Viewed through a high-powered microscope, the alveoli look rather like bunches of grapes, and are covered with tiny blood capillaries (Figure 3.7).

#### HINT

Be careful when interpreting percentages! The *percentage* of a gas in a mixture can vary, even if the actual *amount* of the gas stays the same. This is easiest to understand from an example. Imagine you have a bottle containing a mixture of 20% oxygen and 80% nitrogen. If you used a chemical to absorb all the oxygen in the bottle, the nitrogen left would now be 100% of the gas in the bottle, despite the fact that the *amount* of nitrogen would still be the same. That is why the percentage of nitrogen in inhaled and exhaled air is slightly different.



▲ Figure 3.7 (a) Alveoli and the surrounding capillary network. (b) Diffusion of oxygen and carbon dioxide takes place between the air in the alveolus and the blood in the capillaries.

### HINT

Be careful – students sometimes write ‘The alveolus has *cell walls*’. This statement is not correct – a cell wall is part of a plant cell! The correct way to describe the structure is: ‘The alveolus has a *wall made of cells*’.

Deoxygenated blood is pumped from the heart to the lungs and passes through the capillaries surrounding the alveoli. The blood has come from the respiring tissues of the body, where it has given up some of its oxygen to the cells, and gained carbon dioxide. Around the lungs, the blood is separated from the air inside each alveolus by only two cell layers; the cells making up the wall of the alveolus, and the capillary wall itself. This is a distance of less than a thousandth of a millimetre.

### EXTENSION WORK

The thin layer of fluid lining the inside of the alveoli comes from the blood. The capillaries and cells of the alveolar wall are ‘leaky’ and the blood pressure pushes fluid out from the blood plasma into the alveolus. Oxygen dissolves in this moist surface before it passes through the alveolar wall into the blood.

Because the air in the alveolus has a higher concentration of oxygen than the blood entering the **capillary** network, oxygen diffuses from the air, across the wall of the alveolus and into the blood. At the same time there is more carbon dioxide in the blood than there is in the air in the lungs. This means that there is a diffusion gradient for carbon dioxide in the other direction, so carbon dioxide diffuses the other way, out of the blood and into the alveolus. The result is that the blood which leaves the capillaries and flows back to the heart has gained oxygen and lost carbon dioxide. The heart then pumps the oxygenated blood around the body again, to supply the respiring cells (see Chapter 5).

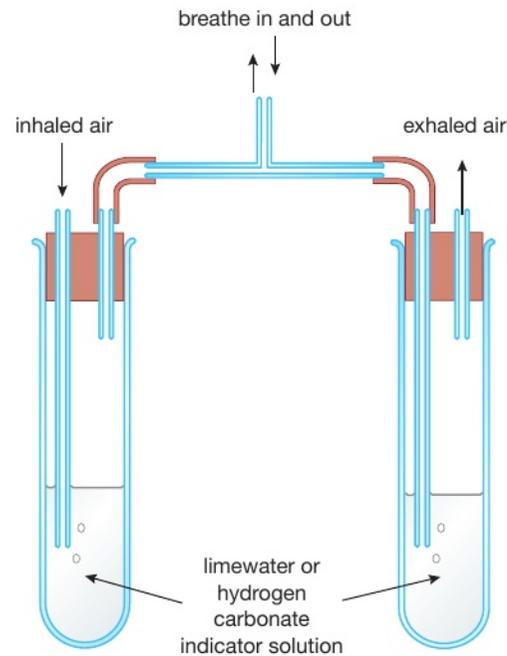


**Safety note:** Wear eye protection and breathe gently; do not blow. A clean mouthpiece must be used for each person.

## ACTIVITY 1

### ▼ PRACTICAL: COMPARING THE CARBON DIOXIDE CONTENT OF INHALED AND EXHALED AIR

The apparatus in Figure 3.8 can be used to compare the amount of carbon dioxide in inhaled and exhaled air. A person breathes gently in and out through the middle tube. Exhaled air passes out through one tube of indicator solution and inhaled air is drawn in through the other tube. If limewater is used, the limewater in the 'exhaled' tube will turn cloudy before the limewater in the 'inhaled' tube. (If hydrogen carbonate indicator solution is used instead, it changes from red to yellow.)



▲ Figure 3.8 Apparatus for Experiment 6.



**Safety note:** Wear suitable footwear for exercising and if doing step-ups use a sturdy secure low box or a PE bench.

## ACTIVITY 2

### ▼ PRACTICAL: AN INVESTIGATION INTO THE EFFECT OF EXERCISE ON BREATHING RATE

It is easy to show the effect of exercise on a person's breathing rate. They sit quietly for five minutes, making sure that they are completely relaxed. They then count the number of breaths they take in one minute, recording their results in a table. They wait a minute, and then count their breaths again, recording the result, and repeating if necessary until they get a steady value for the 'resting rate'.

The person then carries out some vigorous exercise, such as running on the spot for three minutes. Immediately after they finish the exercise, they sit down and record the breathing rate as before. They then continue to record their breaths per minute, every minute, until they return to their normal resting rate.

The table shows the results from an investigation into the breathing rate of two girls, A and B, before and after exercise.

Time from start of experiment (min)	Breathing rate / breaths per min	
	A	B
1	13	13
2	14	12
3	14	12
<b>Rate after 3 min vigorous exercise:</b>		
7	28	17
8	24	13
9	17	12
10	14	12

Plot a line graph of these results, using the same axes for both subjects. Join the data points using straight lines, and leave a gap during the period of exercise, when no readings were taken.

Why does breathing rate need to rise during exercise? Explain as fully as possible. Why does the rate not return to normal as soon as a subject finishes the exercise? (see Chapter 1).

Describe the difference in the breathing rates of the two girls (A and B) after exercise. Which girl is more fit? Explain your reasoning.

## THE EFFECTS OF SMOKING

In order for the lungs to exchange gases properly, the air passages need to be clear, the alveoli need to be free from dirt particles and bacteria, and they must have as big a surface area as possible in contact with the blood. There is one habit that can upset all of these conditions – smoking.

Links between smoking and diseases of the lungs are now a proven fact. Smoking is associated with lung cancer, bronchitis and emphysema. It is also a major contributing factor to other conditions, such as coronary heart disease and ulcers of the stomach and intestine. Pregnant women who smoke are more likely to give birth to underweight babies.

Coronary heart disease will be described in Chapter 5 after you have studied the structure of the heart. Here we will look at a number of other medical conditions that are caused by smoking.

### EFFECTS OF SMOKE ON THE LINING OF THE AIR PASSAGES

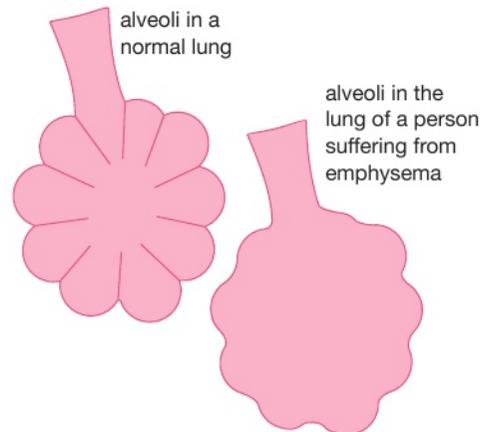
You saw above how the lungs are kept free of particles of dirt and bacteria by the action of mucus and cilia. In the trachea and bronchi of a smoker, the cilia are destroyed by the chemicals in cigarette smoke.

The reduced numbers of cilia mean that the mucus is not swept away from the lungs, but remains to block the air passages. This is made worse by the fact that the smoke irritates the lining of the airways, stimulating the cells to secrete more mucus. The sticky mucus blocking the airways is the source of 'smoker's cough'. Irritation of the bronchial tree, along with infections from bacteria in the mucus, can cause the lung disease **bronchitis**. Bronchitis blocks normal air flow, so the sufferer has difficulty breathing properly.

## EMPHYSEMA

**Emphysema** is another lung disease that kills about 20 000 people in Britain every year. Smoking is the cause of one type of emphysema. Smoke damages the walls of the alveoli, which break down and fuse together again, forming enlarged, irregular air spaces (Figure 3.9).

This greatly reduces the surface area for gas exchange, which becomes very inefficient. The blood of a person with emphysema carries less oxygen. In serious cases, this leads to the sufferer being unable to carry out even mild exercise, such as walking. Emphysema patients often have to have a supply of oxygen nearby at all times (Figure 3.10). There is no cure for emphysema, and usually the sufferer dies after a long and distressing illness.



▲ Figure 3.9 The alveoli of a person suffering from emphysema have a greatly reduced surface area and inefficient gas exchange.



▲ Figure 3.10 Patients with emphysema often need to breathe air enriched with oxygen in order to stay alive.

### EXTENSION WORK

A person who has chronic (long-term) bronchitis and emphysema is said to be suffering from chronic obstructive pulmonary disease, or COPD. COPD is a progressive disease for which there is no cure.

## LUNG CANCER

Evidence of the link between smoking and lung cancer first appeared in the 1950s. In one study, a number of patients in hospital were given a series of questions about their lifestyles. They were asked about their work, hobbies, housing and so on, including a question about how many cigarettes they smoked. The same questionnaire was given to two groups of patients. The first group were all suffering from lung cancer. The second (**Control**) group were in hospital with various other illnesses, but not lung cancer. To make it a fair comparison, the Control patients were matched with the lung cancer patients for sex, age and so on.

When the results were compared, one difference stood out (Table 3.2). A greater proportion of the lung cancer patients were smokers than in the Control patients. There seemed to be a connection between smoking and getting lung cancer.

Table 3.2 Comparison of the smoking habits of lung cancer patients and other patients.

	Percentage of patients who were non-smokers	Percentage of patients who smoked more than 15 cigarettes a day
lung cancer patients	0.5	25
Control patients (with illnesses other than lung cancer)	4.5	13



▲ Figure 3.11 This lung is from a patient with lung cancer.

**DID YOU KNOW?**

People often talk about ‘yellow nicotine stains’. In fact it is the *tar* that stains a smoker’s fingers and teeth. Nicotine is a colourless, odourless chemical.

**DID YOU KNOW?**

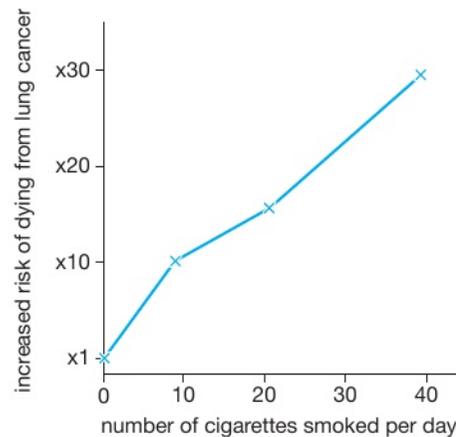
Studies have shown that the type of cigarette smoked makes very little difference to the smoker’s risk of getting lung cancer. Filtered and ‘low tar’ cigarettes only reduce the risk slightly.

Although the results didn’t prove that smoking caused lung cancer, there was a statistically significant link between smoking and the disease: this is called a ‘correlation’.

Over 20 similar investigations in nine countries have revealed the same findings. In 1962 a report called ‘Smoking and health’ was published by the Royal College of Physicians of London, which warned the public about the dangers of smoking. Not surprisingly, the first people to take the findings seriously were doctors, many of whom stopped smoking. This was reflected in their death rates from lung cancer. In ten years, while deaths among the general male population had risen by 7%, the deaths of male doctors from the disease had *fallen* by 38%.

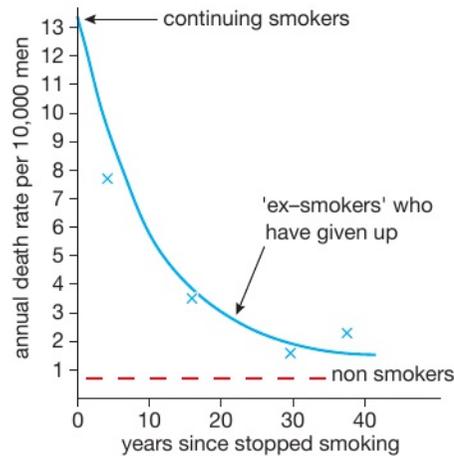
Cigarette smoke contains a strongly addictive drug – **nicotine**. Smoke contains over 7000 chemicals, including; carbon monoxide, arsenic, ammonia, formaldehyde, cyanide, benzene, and toluene. More than 60 of the chemicals are known to cause cancer. These chemicals are called **carcinogens**, and are contained in the tar that collects in a smoker’s lungs. Cancer happens when cells mutate and start to divide uncontrollably, forming a **tumour** (Figure 3.11). If a lung cancer patient is lucky, they may have the tumour removed by an operation before the cancer cells spread to other tissues of the body. Unfortunately tumours in the lungs usually cause no pain, so they are not discovered until it is too late – it may be inoperable, or tumours may have developed elsewhere.

If you smoke you are not *bound* to get lung cancer, but the risk that you will get it is much greater. In fact, the more cigarettes you smoke, the more the risk increases (Figure 3.12).



▲ Figure 3.12 The more cigarettes a person smokes, the more likely it is they will die of lung cancer. For example, smoking 20 cigarettes a day increases the risk by about 15 times.

The obvious thing to do is not to start smoking. However, if you are a smoker, giving up the habit soon improves your chance of survival (Figure 3.13). After a few years, the likelihood of your dying from a smoking-related disease is almost back to the level of a non-smoker.



▲ Figure 3.13 Death rates from lung cancer for smokers, non-smokers and ex-smokers.

## CARBON MONOXIDE IN SMOKE

One of the harmful chemicals in cigarette smoke is the poisonous gas **carbon monoxide**. When this gas is breathed in with the smoke, it enters the bloodstream and interferes with the ability of the blood to carry oxygen. Oxygen is carried around in the blood in the red blood cells, attached to a chemical called **haemoglobin** (see Chapter 5). Carbon monoxide can combine with the haemoglobin much more tightly than oxygen can, forming a compound called **carboxyhaemoglobin**. The haemoglobin will combine with carbon monoxide in preference to oxygen. When this happens, the blood carries much less oxygen around the body. Carbon monoxide from smoking is also a major cause of heart disease (Chapter 5).

If a pregnant woman smokes, she will be depriving her unborn **fetus** of oxygen (Figure 3.14). This has an effect on its growth and development, and leads to the mass of the baby at birth being lower, on average, than the mass of babies born to non-smokers.

### SOME SMOKING STATISTICS

- It is estimated that there are over 1 billion smokers worldwide. In 2014 they consumed 5.8 *trillion* cigarettes.
- Every year nearly 6 million people are killed by tobacco-related illnesses. If the current trend continues, by 2030 this will rise to 8 million deaths per year and 80% of these premature deaths will be in developing countries.
- Smoking causes almost 80% of deaths from lung cancer, 80% of deaths from bronchitis and emphysema, and 14% of deaths from heart disease.
- More than a quarter of all cancer deaths are attributable to smoking. These include cancer of the lung, mouth, lip, throat, **bladder**, kidney, pancreas, stomach, liver and cervix.
- While demand for tobacco has steadily fallen in developed countries like the UK, cigarette consumption is being increasingly concentrated in the developing world.
- 9.6 million adults in the UK smoke cigarettes, 20% of men and 17% of women. However, 22% of women and 30% of men in the UK are now ex-smokers. Surveys show that about two-thirds of current smokers would like to stop smoking.
- It is estimated that worldwide, 31% of men and 8% of women are smokers. Consumption varies widely between different countries, but generally the areas of the world where there has been no change in consumption, or an increase, are southern and central Asia, Eastern Europe and Africa.

- In China alone there are about 350 million smokers, who consume about one-third of all cigarettes smoked worldwide. Large multinational tobacco companies have long been keen to enter the Chinese market.
- In China there are over a million deaths a year from smoking-related diseases. This figure is expected to double by 2025.
- In developing countries, smoking has a greater economic impact. Poorer smokers spend significant amounts of their income on cigarettes rather than necessities like food, healthcare and education.
- Tobacco farming uses up land that could be used for growing food crops. In 2012, 7.5 million tonnes of tobacco leaf were grown on almost 4.3 million hectares of land (an area larger than Switzerland).

Sources: *Action on Smoking and Health (ASH) fact sheets (2015- 2016); ASH research reports (2014-2016)*

## GIVING UP SMOKING

Most smokers admit that they would like to find a way to give up the habit. The trouble is that the nicotine in tobacco is a very addictive drug, and causes withdrawal symptoms when people stop smoking. These include cravings for a cigarette, restlessness and a tendency to put on weight (nicotine depresses the appetite).

There are various ways that smokers can be helped to give up their habit. One method is 'vaping', which involves inhaling a vapour containing nicotine from an electronic cigarette or e-cigarette (Figure 3.15). Other methods use nicotine patches (Figure 3.16) or nicotine chewing gum. They all work in a similar way, providing the smoker with a source of nicotine without the harmful tar from cigarettes. The nicotine is absorbed by the body and reduces the craving for a cigarette. Gradually, the patient reduces the nicotine dose until they are weaned off the habit.

### EXTENSION WORK

You could carry out an Internet search to find out about the different methods people use to help them give up smoking. Which methods have the highest success rate? Is there any evidence that suggests e-cigarettes are not safe?

There are several other ways that people use to help them give up smoking, including the use of drugs that reduce withdrawal symptoms, acupuncture and even hypnotism.

## CHAPTER QUESTIONS

### SKILLS CRITICAL THINKING



More questions on breathing can be found at the end of Unit 2 on page 116.

1 The structures below are found in the human bronchial tree

- |            |                |
|------------|----------------|
| 1. alveoli | 3. bronchioles |
| 2. trachea | 4. bronchi     |

Which of the following shows the route taken by air after it is breathed in through the mouth?

- |                        |                        |
|------------------------|------------------------|
| <b>A</b> 2 → 3 → 4 → 1 | <b>C</b> 2 → 4 → 3 → 1 |
| <b>B</b> 1 → 4 → 3 → 2 | <b>D</b> 4 → 1 → 2 → 3 |



2 Which of the following is *not* a feature of an efficient gas exchange surface?

- |                       |   |
|-----------------------|---|
| <b>A</b> thick walls  | <b>C</b> close proximity to blood capillaries |
| <b>B</b> moist lining | <b>D</b> large surface area                   |

## SKILLS CRITICAL THINKING



- 3 Which row in the table shows the correct percentage of oxygen in atmospheric and exhaled air?

	Atmospheric air / %	Exhaled air / %
A	78	21
B	21	16
C	16	4
D	4	0.04

- 4 Chemicals in cigarette smoke lead to the breakdown of the walls of the alveoli. What is the name given to this disease?

- A bronchitis  
 B emphysema  
 C coronary heart disease  
 D lung cancer



- 5 Copy and complete the table, which shows what happens in the thorax during ventilation of the lungs. Two boxes have been completed for you.

	Action during inhalation	Action during exhalation
external intercostal muscles	contract	
internal intercostal muscles		
ribs		move down and in
diaphragm		
volume of thorax		
pressure in thorax		
volume of air in lungs		



- 6 A student wrote the following about the lungs.

When we breathe in, our lungs inflate, sucking air in and pushing the ribs up and out, and forcing the diaphragm down. This is called respiration. In the air sacs of the lungs, the air enters the blood. The blood then takes the air around the body, where it is used by the cells. The blood returns to the lungs to be cleaned. When we breathe out, our lungs deflate, pulling the diaphragm up and the ribs down. The stale air is pushed out of the lungs. The student did not have a good understanding of the workings of the lungs. Re-write their description, using correct biological words and ideas.

## SKILLS REASONING



- 7 Sometimes, people injured in an accident such as a car crash suffer from a *pneumothorax*. This is an injury where the chest wall is punctured, allowing air to enter the pleural cavity (see Figure 3.1). A patient was brought to the casualty department of a hospital, suffering from a pneumothorax on the left side of his chest. His left lung had collapsed, but he was able to breathe normally with his right lung.

- a Explain why a pneumothorax caused the left lung to collapse.  
 b Explain why the right lung was not affected.  
 c If a patient's lung is injured or infected, a surgeon can sometimes 'rest' it by performing an operation called an *artificial pneumothorax*. What do you think might be involved in this operation?

## SKILLS CRITICAL THINKING



8 Briefly explain the importance of the following.



a The trachea wall contains C-shaped rings of cartilage.



b The distance between the air in an alveolus and the blood in an alveolar capillary is less than 1/1000th of a millimetre.



c The lining of the trachea contains mucus-secreting cells and cells with cilia.



d Smokers have a lower concentration of oxygen in their blood than non-smokers.

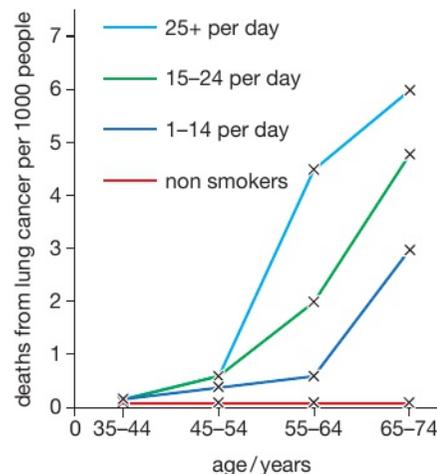
e Nicotine patches and nicotine chewing gum can help someone give up smoking.

f The lungs have a surface area of about 60 m<sup>2</sup> and a good blood supply.

9 Explain the differences between the lung diseases bronchitis and emphysema.

## SKILLS ANALYSIS

10 A long-term investigation was carried out into the link between smoking and lung cancer. The smoking habits of male doctors aged 35 or over were determined while they were still alive, then the number and causes of deaths among them were monitored over a number of years. (Note that this survey was carried out in the 1950s – very few doctors smoke these days!) The results are shown in the graph.



a Write a paragraph to explain what the researchers found out from the investigation.



b How many deaths from lung cancer would be expected for men aged 55 who smoked 25 cigarettes a day up until their death? How many deaths from lung cancer would be expected for men in the same age group smoking 10 a day?

c Table 3.2 (page 43) shows the findings of another study linking lung cancer with smoking. Which do you think is the more convincing evidence of the link, this investigation or the findings illustrated in Table 3.2?

## SKILLS ANALYSIS, PROBLEM SOLVING

## SKILLS ANALYSIS, REASONING

## SKILLS CREATIVITY



11 Design and make a hard-hitting leaflet explaining the link between smoking and lung cancer. It should be aimed at encouraging an adult smoker to give up the habit. You could use suitable computer software to produce your design. Include some smoking statistics, perhaps from an Internet search. However don't use too many, or they may put the person off reading the leaflet!

## 4 FOOD AND DIGESTION

Food is essential for life. The nutrients obtained from it are used in many different ways by the body. This chapter looks at the different kinds of food, and how the food is broken down by the digestive system and absorbed into the blood, so that it can be carried to all the tissues of the body.

### LEARNING OBJECTIVES

- Identify the chemical elements present in carbohydrates, proteins and lipids (fats and oils)
- Describe the structure of carbohydrates, proteins and lipids as large molecules made up from smaller basic units – starch and glycogen from simple sugars, protein from amino acids, and lipids from fatty acids and glycerol
- Investigate food samples for the presence of glucose, starch, protein and fat
- Understand that a balanced diet should include appropriate proportions of carbohydrate, protein, lipid, vitamins, minerals, water and dietary fibre
- Identify the sources and describe the functions of carbohydrate, lipid, protein, vitamins A, C and D, the mineral ions calcium and iron, water, and dietary fibre as components of the diet
- Understand how energy requirements vary with activity levels, age and pregnancy
- Describe the structure and function of the human alimentary canal, including the mouth, oesophagus, stomach, small intestine (duodenum and ileum), large intestine (colon and rectum) and pancreas
- Understand how food is moved through the gut by peristalsis
- Understand the role of digestive enzymes, including the digestion of starch to glucose by amylase and maltase, the digestion of proteins to amino acids by proteases and the digestion of lipids to fatty acids and glycerol by lipases
- Understand that bile is produced by the liver and stored in the gall bladder, and understand the role of bile in neutralising stomach acid and emulsifying lipids
- Understand how the small intestine is adapted for absorption, including the structure of a villus



▲ Figure 4.1 A balanced diet contains all the types of food the body needs, in just the right amounts.

We need food for three main reasons:

- to supply us with a ‘fuel’ for energy
- to provide materials for growth and repair of tissues
- to help fight disease and keep our bodies healthy.

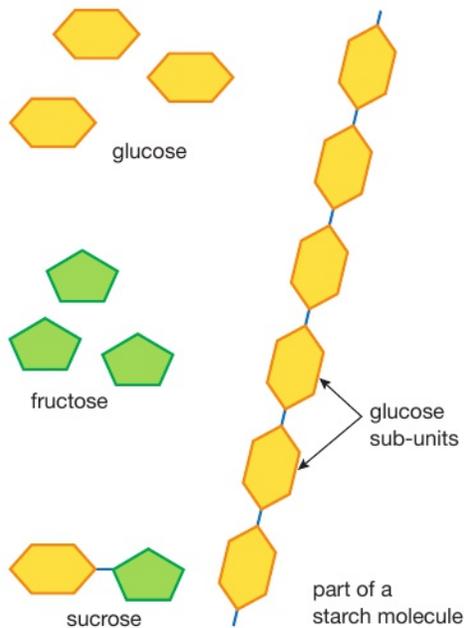
### A BALANCED DIET

The food that we eat is called our diet. No matter what you like to eat, your diet must include the following five groups of food substances if your body is to work properly and stay healthy – **carbohydrates**, **lipids**, **proteins**, **minerals** and **vitamins** – along with **dietary fibre** and water. Food should provide you with all of these substances, but they must also be present in the *right* amounts. A diet that provides enough of these substances and in the correct proportions to keep you healthy is called a **balanced diet** (Figure 4.1). We will look at each type of food in turn, to find out about its chemistry and the role that it plays in the body.

## CARBOHYDRATES

## DID YOU KNOW?

The chemical formula for glucose is  $C_6H_{12}O_6$ . Like all carbohydrates, glucose contains only the elements carbon, hydrogen and oxygen. The 'carbo' part of the name refers to carbon, and the 'hydrate' part refers to the fact that the hydrogen and oxygen atoms are in the ratio two to one, as in water ( $H_2O$ ).



▲ Figure 4.2 Glucose and fructose are 'single sugar' molecules. A molecule of glucose joined to a molecule of fructose forms the 'double sugar' called sucrose. Starch is a polymer of many glucose sub-units.

Carbohydrates only make up about 1% of the mass of the human body, but they have a very important role. They are the body's main 'fuel' for supplying cells with energy. Cells release this energy by oxidising a sugar called **glucose**, in the process called cell respiration (see Chapter 1). Glucose and other sugars belong to one group of carbohydrates.

Glucose is found naturally in many sweet-tasting foods, such as fruits and vegetables. Other foods contain different sugars, such as the fruit sugar called **fructose**, and the milk sugar, **lactose**. Ordinary table sugar, the sort some people put in their tea or coffee, is called **sucrose**. Sucrose is the main sugar that is transported through plant stems. This is why we can extract it from sugar cane, which is the stem of a large grass-like plant. Sugars have two physical properties that you will probably know: they all taste sweet, and they are all soluble in water.

We can get all the sugar we need from natural foods such as fruits and vegetables, and from the **digestion** of starch. Many processed foods contain large amounts of *added* sugar. For example, a typical can of cola can contain up to seven teaspoons (27 g) of sugar! There is hidden sugar in many other foods. A tin of baked beans contains about 10g of added sugar. This is on top of all the food that we eat with a more obvious sugar content, such as cakes, biscuits and sweets.

In fact, we get most of the carbohydrate in our diet not from sugars, but from **starch**. Starch is a large, *insoluble* molecule. Because it does not dissolve, it is found as a storage carbohydrate in many plants, such as potato, rice, wheat and millet. The 'staple diets' of people from around the world are starchy foods like rice, potatoes, bread and pasta. Starch is a polymer of glucose – it is made of long chains of hundreds of glucose molecules joined together (Figure 4.2).

Starch is only found in plant tissues, but animal cells sometimes contain a very similar carbohydrate called **glycogen**. This is also a polymer of glucose, and is found in tissues such as liver and muscle, where it acts as a store of energy for these organs.

As you will see, large carbohydrates such as starch and glycogen have to be broken down into simple sugars during digestion, so that they can be absorbed into the blood.

Another carbohydrate that is a polymer of glucose is **cellulose**, the material that makes up plant cell walls. Humans are *not* able to digest cellulose, because our gut doesn't make the enzyme needed to break down the cellulose molecule. This means that we are not able to use cellulose as a source of energy. However, it still has a vitally important function in our diet. It forms **dietary fibre** or 'roughage', which gives the muscles of the gut something to push against as the food is moved through the intestine. This keeps the gut contents moving, avoiding constipation and helping to prevent serious diseases of the intestine, such as colitis and bowel cancer.

## EXTENSION WORK

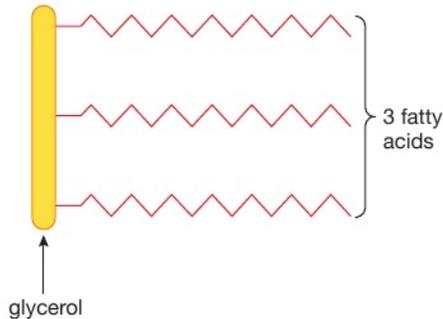
'Single' sugars such as glucose and fructose are called **monosaccharides**. Sucrose molecules are made of two monosaccharides (glucose and fructose) joined together, so sucrose is called a **disaccharide**. Lactose is also a disaccharide, made of glucose joined to another monosaccharide called galactose. Polymers of sugars, such as starch, glycogen and cellulose, are called polysaccharides.

## LIPIDS (FATS AND OILS)

Lipids contain the same three elements as carbohydrates – carbon, hydrogen and oxygen – but the proportion of oxygen in a lipid is much lower than in a carbohydrate. For example, beef and lamb both contain a fat called tristearin, which has the formula  $C_{51}H_{98}O_6$ . This fat, like other animal fats, is a solid at room



▲ Figure 4.3 These foods are all rich in lipids.



▲ Figure 4.4 Lipids are made up of a molecule of glycerol joined to three fatty acids. The many different fatty acids form the variable part of the molecule.

#### KEY POINT

Cholesterol is a substance that the body gets from food such as eggs and meat, but we also make cholesterol in our liver. It is an essential part of all cells, but too much cholesterol causes heart disease.

temperature, but melts if you warm it up. On the other hand, plant lipids are usually liquid at room temperature, and are called oils. Meat, butter, cheese, milk, eggs and oily fish are all rich in animal fats, as well as foods fried in animal fat. Vegetable oils include many types used for cooking, such as olive oil, corn oil and rapeseed oil, as well as products made from oils, such as margarine (Figure 4.3).

Lipids make up about 10% of our body's mass. They form an essential part of the structure of all cells, and fat is deposited in certain parts of the body as a long-term store of energy, for example under the skin and around the heart and kidneys. The fat layer under the skin acts as insulation, reducing heat loss through the surface of the body. Fat around organs such as the kidneys also helps to protect them from mechanical damage.

The chemical 'building blocks' of lipids are two types of molecule called **glycerol** and **fatty acids**. Glycerol is an oily liquid. It is also known as glycerine, and is used in many types of cosmetics. In lipids, a molecule of glycerol is joined to three fatty acid molecules. There are many different fatty acid molecules, which give us the many different kinds of lipid found in food (Figure 4.4).

Although lipids are an essential part of our diet, too much lipid is unhealthy, especially a type called *saturated* fat, and a lipid compound called **cholesterol**. These substances have been linked to heart disease (see Chapter 5).

#### DID YOU KNOW?

*Saturated* lipids (saturated fats) are more common in food from animal sources, such as meat and dairy products. 'Saturated' is a word used in chemistry, which means that the fatty acids of the lipids contain no double bonds. Other lipids are *unsaturated*, which means that their fatty acids contain double bonds. These are more common in plant oils. There is evidence that unsaturated lipids are healthier for us than saturated ones.

## PROTEINS



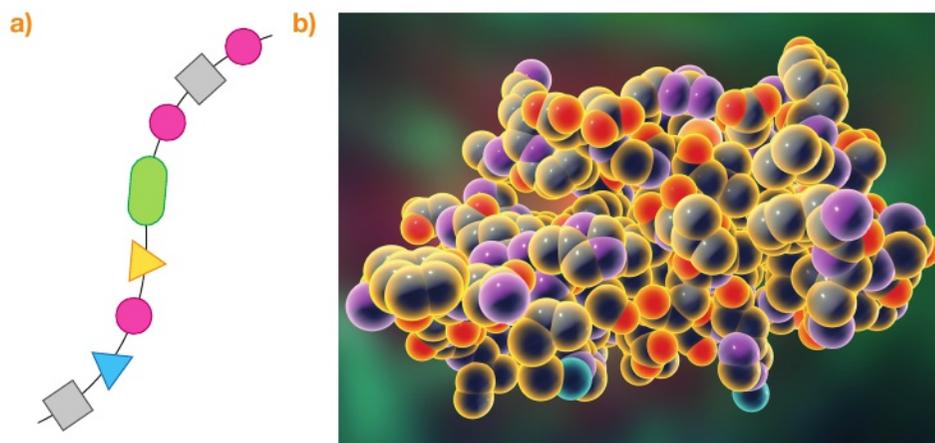
▲ Figure 4.5 This child is suffering from a lack of protein in his diet, a disease called kwashiorkor. His swollen belly is not due to a full stomach, but is caused by fluid collecting in the tissues. Other symptoms include loss of weight, poor muscle growth, general weakness and flaky skin.

Proteins make up about 18% of the mass of the body. This is the second largest percentage after water. All cells contain protein, so we need it for growth and repair of tissues. Many compounds in the body are made from protein, including enzymes.

Most foods contain some protein, but certain foods such as meat, fish, cheese and eggs are particularly rich in it. You will notice that these foods are animal products. Plant material generally contains less protein, but some foods, especially beans, peas and nuts, are richer in protein than others.

However, we don't need much protein in our diet to stay healthy. Doctors recommend a maximum daily intake of about 70g. In more economically developed countries, people often eat far more protein than they need, whereas in many poorer countries a protein-deficiency disease called **kwashiorkor** is common (Figure 4.5).

Like starch, proteins are also polymers, but whereas starch is made from a single molecular building block (glucose), proteins are made from 20 different sub-units called **amino acids**. All amino acids contain four chemical elements: carbon, hydrogen and oxygen (as in carbohydrates and fats) along with nitrogen. Two amino acids also contain sulfur. The amino acids are linked together in long chains, which are usually folded up or twisted into spirals, with cross-links holding the chains together (Figure 4.6).



▲ Figure 4.6 (a) A chain of amino acids forming part of a protein molecule. Each shape represents a different amino acid. (b) A computer model of the protein insulin. This substance, like all proteins, is made of a long chain of amino acids arranged in a particular order and folded into a specific shape.

### EXTENSION WORK

Humans can make about half of the 20 amino acids that they need, but the other 10 have to be taken in as part of the diet. These 10 are called essential amino acids. There are higher amounts of essential amino acids in meat, fish, eggs and dairy products. If you are a vegetarian, you can still get all the essential amino acids you need, as long as you eat a varied diet that includes a range of different plant materials.

The *shape* of a protein is very important in allowing it to carry out its function, and the *order* of amino acids in the protein decides its shape. Because there are 20 different amino acids, and they can be arranged in any order, the number of different protein structures that can be made is enormous. As a result, there are thousands of different kinds of proteins in organisms, from structural proteins such as collagen and keratin in skin and nails, to proteins with more specific functions, such as enzymes and haemoglobin.

### MINERALS

All the foods you have read about so far are made from just five chemical elements: carbon, hydrogen, oxygen, nitrogen and sulfur. Our bodies contain many other elements that we get from our food as ‘minerals’ or ‘mineral ions’. Some are present in large amounts in the body, for example calcium, which is used for making teeth and bones. Others are present in much smaller amounts, but still have essential jobs to do. For instance our bodies contain about 3g of iron, but without it our blood would not be able to carry oxygen. Table 4.1 shows just a few of these minerals and the reasons they are needed.

Table 4.1 Some examples of minerals needed by the body.

Mineral	Approximate mass in an adult body / g	Location or role in body	Examples of foods rich in minerals
calcium	1000	making teeth and bones	dairy products, fish, bread, vegetables
phosphorus	650	making teeth and bones; part of many chemicals, e.g. DNA and ATP	most foods
sodium	100	in body fluids, e.g. blood	common salt, most foods
chlorine	100	in body fluids, e.g. blood	common salt, most foods
magnesium	30	making bones; found inside cells	green vegetables
iron	3	part of haemoglobin in red blood cells, helps carry oxygen	red meat, liver, eggs, some vegetables, e.g. spinach